

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24

ILLINOIS POLLUTION CONTROL BOARD  
June 16, 2006

IN THE MATTER OF )  
 )  
PROPOSED NEW 35 ILL. ADM. CODE) R06-25  
225 CONTROL OF EMISSIONS FROM) (Rulemaking - Air)  
LARGE COMBUSTION SOURCES )  
(MERCURY) )

TESTIMONY OF DR. GERALD KEELER  
PART II

BEFORE MARIE E. TIPSORD  
HEARING OFFICER

The testimony of Dr. Gerald Keeler, a witness called in the rulemaking proceeding before the Illinois Pollution Control Board taken on June 16, 2006, at 9:00 a.m., at the offices of the Environmental Protection Agency, Springfield, Illinois, before Holly A. Schmid, Notary Public and Certified Shorthand Reporter, CSR No. 084-98-254587 for the State of Illinois.

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24

A P P E A R A N C E S

MEMBERS OF THE ILLINOIS POLLUTION CONTROL BOARD:  
Ms. Marie E. Tipsord, Hearing Officer;  
Dr. G. Tanner Girard, Board Member;  
Ms. Andrea S. Moore, Board Member;  
Mr. Anand Rao, Board Staff;  
Mr. Thomas Johnson, Board Staff;  
Mr. Tim Fox, Board Staff;  
Mr. Nicholas Melas, Board Staff;  
Ms. Alisa Liu, Board Staff.

COUNSEL FOR THE ILLINOIS  
ENVIRONMENTAL PROTECTION AGENCY:  
Mr. Charles Matoesian;  
Ms. Gina Roccaforte;  
Mr. John Kim;  
Mr. Richard Ayres;

COUNSEL FROM SCHIFF-HARDEN  
Ms. Kathleen Bassi;  
Mr. Stephen Bonebrake;  
Mr. Sheldon Zabel;  
Mr. Jim Ingram, Dynegey, Inc.

COUNSEL FROM JENNER & BLOCK  
Mr. Bill Forcade;  
Ms. Katherine Rahill.

COUNSEL FROM McGUIRE-WOODS:  
Mr. James Harrington;  
Mr. David Rieser.

COUNSEL FROM THE CHICAGO LEGAL CLINIC  
Mr. Keith I. Harley

COUNSEL FROM MIDWEST GENERATION  
Mr. Basil G. Constantelos

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24

IDENTIFICATION

PG.

Exhibit 29:	5
Exhibit 30:	5
Exhibit 31:	95
Exhibit 32:	130

1                   MADAM HEARING OFFICER: Good morning,  
2 everyone. Welcome back. This is day five, and we're  
3 making some good progress.

4                   My name is Maria Tipsord. I'm the  
5 hearing officer in this proceeding in No. R06-25. My  
6 opening will be fairly truncated, since I don't see any  
7 new faces.

8                   I just want to introduce the panel to  
9 you today. Board member Dr. G. Tanner Girard; Board  
10 Member Anand Rao; the far right Nicholas Melas. To my  
11 far left, Tom Johnson, also a board member. Our  
12 technical unit today is represented by Anand Rao. Tim  
13 Fox is with us again today, and Matt Reed is joining us  
14 today with the board staff.

15                   Before we go back to Dr. Keeler's  
16 testimony, Mr. Rieser, I believe you were going to have  
17 some exhibits for me this morning.

18                   MR. RIESER: Why, yes, I do. We have  
19 copies of the preliminary results of Steubenville. I'm  
20 giving one to Mr. Harley, right off the bat, "The  
21 Preliminary Results for Steubenville Mercury Deposition  
22 Source Apportionment Study" from Tim Opelt, April 27,  
23 2005. I don't recall what number you have that at.

24                   MADAM HEARING OFFICER: We're going to

1 mark that as Exhibit 29, if there's no objection.

2 Seeing none, that will be Exhibit 29.

3 (Exhibit 29 was admitted.)

4 MR. RIESER: Then the second one is  
5 "The Ambient Measurements to Support Coal Combustion  
6 Emission Research" October 21, Beijing, China presented  
7 by Matthew S. Landis, 2005. Again, we are presenting a  
8 copy to Mr. Harley. We'll mark that as Exhibit No. 30,  
9 if there's no objection. Seeing none, we will mark that  
10 as Exhibit No. 30.

11 (Exhibit 30 was admitted.)

12 MS. BASSI: What was 29?

13 MADAM HEARING OFFICER: 29 is the  
14 preliminary results. And "The Ambient Measurement" is  
15 the title page on the second one. That will be Exhibit  
16 30. Thirty is the Beijing study; 29 is the U.S. EPA  
17 preliminary. Have we got the exhibits straightened out?  
18 Has everybody got 29 and 30? I believe we are ready to  
19 start with Question No. 10 for Dr. Keeler.

20 DR. KEELER: Question No. 10: "In  
21 using the source-receptor approach to determine the  
22 source of wet deposition mercury, is it accurate that  
23 you determine the source by analyzing for other  
24 constituents associated with those sources? A, does

1 that analysis demonstrate the proximity of those  
2 sources? B: Does this analysis differentiate among  
3 different types of coal used?" I assume you're asking  
4 source receptor modeling approach not -- in using the  
5 source-receptor approach.

6 MR. RIESER: Correct.

7 DR. KEELER: If you're referring to  
8 the multivarious statistical analysis approach, it does  
9 not demonstrate the proximity of the sources, so the  
10 answer to A is no. If you're asking about the combined  
11 hybrid receptor modeling approach, the answer is yes, so  
12 just to differentiate --

13 MR. RIESER CONTINUES:

14 Q. Just for the record I note that the term  
15 source-receptor study is the one that you used in your  
16 testimony. For example, in this second paragraph of the  
17 conclusion, we talked about, on page 5, that says,  
18 "Source receptor studies have recently been completed  
19 that indicate the coal-fired utilities contributed,  
20 approximately, 70 percent of the mercury," so it wasn't  
21 a term I made up. It was something that I thought I  
22 took directly from your testimony.

23 A. Right. All the methods that we use are  
24 looking for source-receptor relationships, and I was

1 just asking a clarifying question. That's all. It's a  
2 correct use of the term.

3 Q. Thank you.

4 DR. KEELER: Question No. 11 -- just  
5 so it's clear, B, the answer was no. "In your  
6 testimony, you state that coal combustion was clearly  
7 dominant, in terms of explaining the mercury deposition,  
8 approximately, 70 percent. What is the basis for this  
9 figure?" Do you want me to answer this again?

10 MR. RIESER: Yes, please.

11 DR. KEELER: This was the average of  
12 the results of the PMF and the Unmixed models.

13 MR. RIESER CONTINUES:

14 Q. Of the two years, correct?

15 A. For the two years.

16 Q. 2003 and 2004?

17 A. Right, for the two-year combined analysis.

18 Q. There were differences in meteorology  
19 between the two years, were there not?

20 A. Yes, there was. I said the deposition was  
21 quite different between the two years.

22 Q. And in 2004, if I recall your testimony,  
23 was dominated by four major precipitation events. Is  
24 that correct?

1           A.     Actually, I don't recall saying four, but  
2 if that's what I said.

3           Q.     I think you said "several" I'm sorry, but  
4 it was dominated by several large precipitation events?

5           A.     Yes, that's correct.

6           Q.     And in what way was it dominated by those  
7 events?

8           A.     Those were very large deposition totals  
9 relative to the total amount for the entire year.

10          Q.     And what does that mean with respect to  
11 your findings, if any?

12          A.     Well, in one respect what it means is that  
13 for a model, such as a deterministic model, if a missed  
14 accurately calculating the deposition for that one  
15 event, it will significantly alter their source-receptor  
16 relationship and the ultimate apportionment of  
17 whatever came from whatever source.

18          Q.     I'm sorry. What does its mean for the  
19 findings of the receptor study, the one you performed?

20          A.     Not much. I mean, it's just another  
21 example. I was highlighting the fact that it can have a  
22 few single events that can lead to very large  
23 deposition, and it's important in a sense of if one is  
24 trying to compare the results of a CMAQ or TEAM, or that



1 type of a model to a receptor modeling that those single  
2 events are very, very important, in terms of the overall  
3 deposition to that location for the entire year, and so  
4 not accurately predicting them is a major error, so that  
5 was the purpose in my presentations and in my comment.

6 Q. Does it also suggest that source-receptor  
7 done in a year with -- let me ask this, did you consider  
8 -- and I do see in your testimony it's "a few large  
9 precipitation events," I'm sorry. Is it your belief  
10 that these large precipitation events were unusual with  
11 the relationship to the typical weather in Steubenville?

12 A. They are not unusual in the sense that we  
13 see these large events, typically at every site that we  
14 measure. The sites in Michigan, the sites in Vermont,  
15 the sites in Florida, we'll see a couple of very large  
16 events that contribute a fairly large percentage of the  
17 total deposition in all the sites that we make  
18 measurements at.

19 Q. Is it accurate that several of these  
20 precipitation events you observed in Steubenville were  
21 results of hurricanes that came up from the Gulf of  
22 Mexico?

23 A. Two of the events were the result of  
24 cyclones that came up from the Gulf. They were the

1           remnants of hurricanes and provided very large amounts  
2           of precipitation.

3           Q.     Had you done the study in 2004, and only  
4           had the 2004 data, would that have skewed the data  
5           higher than the usual data that you would see for  
6           Steubenville because of the large precipitation events?

7           A.     No.

8           Q.     In what way would it not?

9           A.     It turns out that, for those large  
10          precipitation events, the source-receptor model  
11          underestimated the total deposition for those events,  
12          and in fact, underestimates them quite a bit.  Extreme  
13          events in statistical analyses tend to be smoothed out,  
14          so those couple big events, in terms of the deposition  
15          -- they were over a microgram per square meter per event  
16          -- were underestimated, and therefore, would not have  
17          overestimated or given more of an emphasis to those  
18          events.

19          Q.     But you stated in your testimony that, at  
20          least, one of those events added 8 percent of the total  
21          mass of mercury to the amount of mercury observed for  
22          the year 2004.

23          A.     That's correct.

24          Q.     And you don't consider that a significant

1 additional amount based on that one storm?

2 A. No. The point of my testimony was that it  
3 does -- 8 percent is a significant additional amount,  
4 but you asked the question whether that had a  
5 significant impact on the results of our source-receptor  
6 modeling, and I said no.

7 Q. I understand. I see where the four event  
8 comes from. I'm sorry, Madam Hearing Officer. Which  
9 was the Beijing?

10 MADAM HEARING OFFICER: The Beijing  
11 study is 30?

12 MR. RIESER CONTINUES:

13 Q. Thank you. If you turn to Exhibit 30,  
14 there is an "EPA PMF Estimated, versus Measured Mercury  
15 Deposition."

16 A. Yes, I see that.

17 Q. And it's a graph comparing mercury  
18 deposition in events, and there are four peaks, at  
19 least, four large peaks that are larger than the other  
20 peaks in that graph.

21 A. Correct.

22 Q. Do those represent the significant  
23 precipitation events that you discussed?

24 A. Those are, yes.

1 Q. So would you agree that, in order for the  
2 source-receptor -- excuse me -- receptor modeling to --  
3 strike that. Let's go on to the next question.

4 DR. KEELER: "What range of values is  
5 represented by this approximate value?" An approximate,  
6 that is. I'm sorry. What did I say?

7 MR. RIESER: I thought I heard you say  
8 an approximate.

9 DR. KEELER: "An approximate" it says.  
10 I'm sorry, if I mumbled my words. The PMF and Unmixed  
11 numbers, if I said I think PMF came out to be,  
12 approximately, 70, might have been 69, and Unmix was 74,  
13 and so that's where the number of, approximately, 70.  
14 That wasn't meant to be a mathematically-rounded number,  
15 by any means. It was supposed to be an approximation.

16 MR. RIESER CONTINUES:

17 Q. I asked this, I suppose, indeed, but is  
18 there a plus or minus range associated with that?

19 A. Yeah. As I said yesterday, I believe the  
20 uncertainty, if you do it the same way for both of the  
21 models, is around 15 percent.

22 Q. Thank you.

23 DR. KEELER: The question is asking,  
24 "What is the purpose of expressing this as an

1 approximate value?" For the purpose of simplicity, and I  
2 have already answered what level of uncertainty to place  
3 on this figure, which is the 15 percent. Question 12:  
4 "You state that `a meteorological analysis corroborates  
5 that a substantial amount of the mercury deposition  
6 found at the Steubenville site was due to local and  
7 regional sources."

8 MR. RIESER CONTINUES:

9 Q. Let me stop you. You use the term  
10 "corroborate." What findings do you have, aside from  
11 the meteorologic, that indicate a substantial amount of  
12 mercury was due to local and regional sources?

13 A. Well, if one takes the emission  
14 inventories for the region for mercury, and looks at the  
15 amount of mercury and other constituents that are there,  
16 one can then see that the largest contributor in the  
17 emissions inventory in that region is actually  
18 coal-fired utilities, so again, that's what I'm saying  
19 is we, again, find that this matches up with what the  
20 EPA emission inventory tells us.

21 Q. The emissions inventory is of emissions,  
22 not deposition, correct?

23 A. It is emissions inventory. It's the  
24 actual amount of emissions estimated from the source.

1 Q. So the emissions inventory doesn't, in and  
2 of itself, tell you what the deposition is, does it?

3 A. No, it does not, unless you use a  
4 deterministic model to model from emissions to  
5 deposition?

6 Q. As I understand it, you did not use a  
7 deterministic model?

8 A. No, I did not.

9 Q. And so in what way did you arrive at the  
10 finding through your study that a substantial amount of  
11 mercury deposition found at the Steubenville site was  
12 due to local and regional sources?

13 A. When one does a source-receptor modeling,  
14 one will often take the data, do the analysis, determine  
15 an estimate of what a local source contributions and  
16 regional source contributions are and then use the local  
17 emissions inventory to what we call -- it's not a  
18 validation. It's a comparison to the results that you  
19 find. It doesn't make sense, in terms of what was  
20 emitted in the region, to what you actually predicted.  
21 This is a very common practice. It's done in a lot of  
22 receptor modeling results. For example, if you estimate  
23 that diesel emissions are an important emissions are  
24 important and go to an area, and there are no diesel

1 trucks estimated to have emissions in that area, then  
2 you find your results very unsatisfying.

3 Q. What is it about the emissions inventory  
4 that allows you to make a qualitative statement with  
5 respect to the deposition in the area, as you say, it's  
6 a substantial amount that was due to local and regional  
7 sources?

8 A. Are you asking me about the emissions  
9 inventory?

10 Q. I'm asking you what about the emissions  
11 inventory allows you to make the statement that a  
12 substantial amount of mercury deposition found at the  
13 Steubenville site was due to local and regional sources?

14 A. We have observational evidence that shows  
15 us that, when we have high concentrations of reactive  
16 mercury in the atmosphere, we would expect that to be  
17 removed fairly close by precipitating systems. Taking  
18 this understanding of the processes, one can then take  
19 and use that understanding when looking at, okay, here  
20 are emissions that are, to a large fraction, greater  
21 than 50 percent as estimated by the utilities to a  
22 common form of reactive mercury. When one goes and  
23 looks at when and where the precipitation fell, one can  
24 then make an estimate that this mercury would have been

1 removed through wet deposition in that area.

2 Q. How do you know what the amount of  
3 reactive gaseous mercury is emitted by the sources  
4 around Steubenville?

5 A. It's in the emissions inventory.

6 Q. The amount of reactive gaseous mercury is  
7 in the inventory?

8 A. That's right. They provide an estimate of  
9 the percentage of reactive mercury, particulate mercury  
10 and elemental mercury that's emitted from each source.

11 Q. Did you do anything to -- the emissions  
12 inventory is from 1999. Is that correct?

13 A. That's correct.

14 Q. Did you do anything to evaluate whether  
15 that emissions inventory was still accurate as of the  
16 time you did your study?

17 A. We only have the data that is presented to  
18 us, so the answer would be no. There was no follow-up.  
19 We can only use the information that is provided to us,  
20 so the same information that all the modelers use.

21 Q. Could you have taken steps to -- strike  
22 that. Put it this way, yesterday, when I asked you  
23 about the sources that were nearby, you said that you  
24 didn't know what they were, and you directed us to the



1 EPA map.

2 A. I don't think that's what I said, that I  
3 didn't know what they were. You asked me how many,  
4 specifically, and how much power was produced by the  
5 plants, and I believe I answered I didn't have that  
6 number in front of me.

7 Q. Did you have --

8 A. I know exactly where the sources are, not  
9 just the coal-fired utility, but all of the other  
10 sources in the vicinity.

11 Q. Did you have that number in front of you  
12 when you performed your study?

13 A. We combine all the analysis tools that we  
14 have including the emissions inventories, the maps  
15 showing the locations of all the sites, how much of the  
16 mercury is emitted from all those, and trying to  
17 understand the results of our modeling, yes.

18 Q. Now, of course, we are limited in that we  
19 don't actually have a copy of your study, but were there  
20 maps showing the locations of the power plants, the coal  
21 used, and then megawattage included as part of your  
22 study?

23 A. The publication has a map showing all the  
24 locations of the coal-fired utilities based on the 1990

1 information that was provided -- 1999, excuse me. I  
2 misspoke. And I would have to go back and remember if  
3 in the manuscript version that we submitted, whether it  
4 had anything showing the actual amount of mercury that  
5 was emitted from each stack. I don't recall that, but  
6 it does show, geographically, where the plants were  
7 located.

8 Q. So your report identifies specific power  
9 plants as contributing to the mercury levels that you  
10 observed in Steubenville. Is that correct?

11 A. No. As I mentioned yesterday, the  
12 receptor modeling approach does not allow you to  
13 identify a particular power plant. It identifies source  
14 contributions from coal combustion.

15 Q. Then what was the point of including a map  
16 of certain power plants in your study?

17 A. It's very important when you're publishing  
18 any type of paper in international journals to show  
19 people the location of where you are studying and the  
20 major emissions sources in that vicinity. This is  
21 common practice in the scientific literature.

22 Q. So the point was just to show the  
23 location. Did you have other features, such as items of  
24 historical interest in Steubenville?

1           A.     This was a scientific peer-reviewed  
2 literature paper. It's not a historical overview of the  
3 area of Steubenville.

4           Q.     I'm trying to understand why, if the point  
5 of a receptor study, such as the one you performed, is  
6 that it's not, A, can't use it to identify specific  
7 sources, and B, it's not important to know what those  
8 sources are, I'm trying to understand why you would  
9 identify specific sources within the study, itself.

10          A.     Again, you have misinterpreted and  
11 misquoted what I said. A, I did not say it was not  
12 important to know are the individual sources are. B, I  
13 didn't say that individual sources are not important,  
14 and I've been working in Eastern Ohio, Western  
15 Pennsylvania air shed understanding the sources of, not  
16 just mercury, but other compounds for over 20 years. I  
17 have driven that area myself more hours than I can tell  
18 you. I have spent an enormous amount of time on the  
19 ground. I have actually gone around and photographed the  
20 sources. Identified that, in fact, they were in the  
21 locations that were given because, in the early days, we  
22 were given locations for plants that actually didn't  
23 coincide with where the plants were located, so we  
24 actually went and verified that they were there, so I

1        have a long historical context, in terms of the research  
2        I have been doing in that area. I published my first  
3        source receptor paper on aerosols and their sources in  
4        the Southwestern Pennsylvania area in 1986, so this is  
5        not something that is new to me, and I think, from over  
6        20 years worth of scientific experience, I believe I  
7        have a good sense for what information is important to  
8        provide in a scientific peer-reviewed paper and what's  
9        not, and so I'm not sure why you keep saying I'm  
10       referring to these things as being unimportant.

11                Q.        Well, in your testimony, on page four, in  
12       the paragraph that begins, "Multivariant statistical  
13       receptor models have been successfully used to apportion  
14       sources of mercury deposited in South Florida and  
15       elsewhere." What you say after that in the next  
16       sentence is, "These techniques have the advantage of not  
17       requiring prior measurements of source profiles or  
18       emission inventories." Similarly, in your report, which  
19       is attached as Exhibit B, and I believe it's on page 18,  
20       again, in a paragraph that begins with the statement,  
21       "Regarding multivariant statistical receptor models,"  
22       there's a sentence right before the end of the paragraph  
23       that says, "Again, both techniques referring to  
24       statistical techniques such as Unmix and deposit matrix

1 factorization, both techniques have the advantage of not  
2 requiring prior measurements and source profiles or  
3 emission inventories." So since you identify an  
4 important, positive point that it doesn't require -- the  
5 studies you performed don't require emissions  
6 inventories, I'm having trouble understanding what role  
7 they have to play in the findings that you make, and  
8 obviously, I'm hampered that I don't have the actual  
9 report in front of me.

10 A. I will try to say it again in a different  
11 way, so perhaps it's more clear. Receptor modeling  
12 allows you to estimate the source contributions by  
13 source category for the pollutants that you're looking  
14 at. It does not identify the specific stack or source,  
15 a specific company that emitted the pollutant. You  
16 asked the question about corroboration, and this is how  
17 we got into discussing emissions and emissions  
18 inventories, what we call "source reconciliation." If  
19 one does a receptor modeling study, one that comes to  
20 mind is one Glenn Kass did in the early 70's looking at  
21 the sources in the Los Angeles Basin. He was looking at  
22 the source of L and L carbon (phonetic) and basically,  
23 did a receptor modeling very similar to what we done in  
24 this study, and estimated the contributions to L and L

1 carbon and then went, as a corroborating piece of  
2 information, took the inventory that estimated, from a  
3 source perspective, how much carbon came from all the  
4 sources in LA to see if he could get an approximate  
5 balance. What goes up, has to come down. If you  
6 estimate an overabundance of material that you can't  
7 corroborate with observations or with emissions  
8 estimates, then it gives you some idea that the  
9 emissions are not correct or that there are other  
10 sources contributing that perhaps the emission inventory  
11 didn't conclude, so this is a standard practice in the  
12 field, but it does not require -- receptor modeling does  
13 not require the use of emissions data.

14 Q. Did you quantify the amount of mercury  
15 deposition that you expected to find in Steubenville  
16 based on the emissions inventory?

17 A. No, sir.

18 Q. In what way did the review of the  
19 emissions inventory, then, corroborate the finding that  
20 a substantial amount of mercury deposition found at the  
21 Steubenville site was due to local and regional sources?

22 MADAM HEARING OFFICER: With all due  
23 respect, I think that's where this all started. I think  
24 he answered that question in the beginning.

1 MR. RIESER: Well, I'm really limited  
2 here because I don't think he did answer the question.  
3 He said he looked at the inventory and that supported  
4 the finding that a substantial amount of mercury  
5 deposition found at the Steubenville site was due to  
6 local and regional studies because that finding is  
7 corroborated by other information I have to assume  
8 that's a finding, and again, I don't have the report in  
9 front of me.

10 MADAM HEARING OFFICER: You have made  
11 that point several times this morning, as well.

12 MR. RIESER: Right, but the point  
13 being says he looked at the emissions inventory. We  
14 don't know what plants he looked at. We don't know what  
15 particular things he found. We don't know whether this  
16 was a qualitative or quantitative finding of his.

17 MADAM HEARING OFFICER: I understand  
18 that, but he has also repeatedly stated that the  
19 receptor modeling does not look at individual plants.  
20 It looks at a source category. Am I characterizing that  
21 correctly?

22 DR. KEELER: Correct.

23 MR. MATOESIAN: Yes, and he's answered  
24 that several times, category of sources you can trace

1 back to.

2 MR. RIESER: Thank you.

3 DR. KEELER: Just for the record, I  
4 did state that we use the emissions inventory from the  
5 1999 modeling that EPA did, the same inventory that was  
6 used in the CAMR Rule, and we looked at all the  
7 emissions for the entire Midwest, and actually, for all  
8 the entire Eastern United States, so --

9 MR. RIESER:

10 Q. What I would like, then, is a list of the  
11 plants that Dr. Keeler identified as being the plants  
12 that he looked at the inventory for to identify the  
13 amount of mercury that corroborated this finding.

14 A. It's the entire list that's in the CAMR  
15 Rule, so if you want the entire listing, you can get  
16 that on the EPA website.

17 MADAM HEARING OFFICER: Once again,  
18 Dr. Keeler, when you start telling us to get stuff, that  
19 means the Agency needs to provide it to us, if you have  
20 not already. We need that as an exhibit.

21 MR. RIESER CONTINUES:

22 Q. I'm sorry, the entire list of all power  
23 plants in the country?

24 A. That's the list we used and we looked at



1 the plants east of the Mississippi.

2 Q. Did you look at a specific range of plants  
3 for this study, all the plants in the Mississippi --

4 A. We did not look at specific plants, as I  
5 said earlier. We looked at the region.

6 Q. Well, I'm --

7 A. We didn't identify -- we did not identify  
8 individual plant contributions that are in the receptor  
9 modeling, or in terms of trying to understand the  
10 contribution of a particular plant. That's not in the  
11 scope of the work that we did.

12 MR. RIESER: I don't think it's unfair  
13 to ask what plants, to get a list of the plants.

14 MADAM HEARING OFFICER: I have already  
15 asked them.

16 MR. RIESER: I'm hearing "region" and  
17 "list." I'm not hearing, "I looked at these plants,"  
18 and I'm assuming there is a list.

19 MR. MATOESIAN: He stated this --

20 MR. RIESER: I'm not done talking.  
21 I'm assuming there is a list of plants that he looked at  
22 because he describes, "We looked at these RGM's, and we  
23 calculated the RGM's with the" -- blah, blah blah, and  
24 that's what we need to corroborate, the meteorological

1 findings, so I would like a list of those plants.

2 MR. MATOESIAN: I believe you directed  
3 us to provide that and as he stated, it was a list of  
4 plants east of the Mississippi, I believe.

5 DR. KEELER: Right.

6 MS. BASSI CONTINUES:

7 Q. Sorry to jump into this fray, but please  
8 correct me if I'm wrong, and that's my question. I  
9 thought, at one point, you said that, in your  
10 manuscript, or in your published data, or some place,  
11 there was a map on which you identified the plants that  
12 were in the region or in the vicinity or in the local  
13 area, or however you defined it, but it sounded to me as  
14 if you were saying that there were certain power plants,  
15 perhaps, other types of sources that were on a map  
16 around the Steubenville area that is plotted on a map  
17 that you have identified as dots on a map. Is that  
18 correct?

19 A. That's correct. In the submitted version  
20 of the paper we submitted, a figure, which had a copy --  
21 I mean had a spot showing the location of the  
22 Steubenville site, and then had within -- I would say it  
23 showed the greater Steubenville area, including most of  
24 Ohio, half of Pennsylvania, down to the south of it, up

1 to the north. It didn't include a large area. It was  
2 really meant to show where Steubenville is, but it did  
3 include a point showing a major source of mercury from  
4 the '99 inventory plotted on that.

5 Q. Then somehow that got expanded to the  
6 entire emissions inventory used in the CAMR?

7 A. Because there are two different questions.  
8 One was asking a question about figures, and one was  
9 asking what we look at when we look at the emissions,  
10 and I was trying to be clear about that.

11 Q. So when you looked at the emissions to  
12 corroborate your findings from your study, you were  
13 looking at more than just what was plotted on your map.  
14 Is that correct?

15 A. That is correct.

16 MR. ZABEL CONTINUES:

17 Q. Maybe this is obvious, but it seems to me  
18 we have been dancing around the question of why we can't  
19 get a copy of the draft report.

20 A. It's not a report. It's a manuscript that  
21 was submitted to a scientific journal that's in the  
22 review process. We've already been given reviews, and  
23 we're in the process of finalizing that paper. The  
24 paper will be provided to everyone, once the paper is

1 sent for publication. One other thing that is very  
2 difficult today, especially when doing environmental  
3 research, is to maintain the integrity of the scientific  
4 research publication process, and one of the reasons why  
5 we didn't want to make presentations and do briefings in  
6 front of the EPA, until we had our final analysis was  
7 because these results, then, get misinterpreted. People  
8 that don't do scientific research often don't understand  
9 the process that science is -- you learn as you go. You  
10 make corrections and improvements to the scientific  
11 research that you are doing, and that, at times, if you  
12 don't have the data that you would like at hand, you get  
13 that data, and people provide new data. You submit your  
14 information and you refine your analysis, and if  
15 versions of publications are going around prior to you  
16 actually submitting it to the peer-reviewed literature,  
17 then the whole process becomes -- basically, the  
18 integrity of the process disappears.

19 Q. I'm sorry. My question wasn't answered.

20 A. I'm saying that this hearing has come at a  
21 time where the paper has not come out in press, yet, so  
22 that's the reason why the paper hasn't been provided.

23 Q. I perfectly understand that. I want to  
24 know is there some legal reason why we cannot have that

1 document as it exists today? They are asking this Board  
2 to rely on what you're saying is in that paper. We have  
3 no chance to judge that. If I were in litigation, I  
4 could get it in Discovery. I could order it produced.  
5 They want this Board to rely on what Dr. Keeler is  
6 telling us. I have no reason to think he's not telling  
7 us the truth, but I can't check it. I don't know of a  
8 legal reason -- if there is one, counsel can tell me --  
9 why we can't have copy of that draft.

10 MADAM HEARING OFFICER: I actually  
11 believe that yesterday Dr. Keeler invoked intellectual  
12 property.

13 MR. ZABEL: Then he can have it under  
14 confidentiality. We won't publish it. That's not the  
15 point.

16 DR. KEELER: I'm not sure how to  
17 answer you. I just know that I know that, in terms of  
18 the scientific peer review process, this is not  
19 appropriate and --

20 DR. GIRARD: Could I ask a question.  
21 You say it's been reviewed. Have you already submitted  
22 the revised paper to the journal?

23 DR. KEELER: No, we have not. It's --  
24 right now it's under going, again, another level of

1 review at the Agency, so it's not a finalized document,  
2 yet, and that's the main reason why -- if I hadn't  
3 submitted the final version, I would have no problem  
4 with sharing that, but again, this is -- it's one of  
5 these thing that it's just not done this way.

6 MR. RAO: And based on the comments  
7 you have received from this peer-review committee, does  
8 it change your results or conclusions in any way?

9 DR. KEELER: No, it does not.

10 MS. MOORE: Is the peer view committee  
11 looking at it now within the EPA?

12 DR. KEELER: Yes. It's an internal  
13 review.

14 MS. MOORE: Given the fact that your  
15 findings are U.S. EPA excuse me -- are a little  
16 different than the direction that the U.S. EPA has  
17 thought to go in regard to mercury, don't you think  
18 there would be some chance that the release of this  
19 paper might be held for some period of time for further  
20 and further and further peer review?

21 DR. KEELER: I don't believe they,  
22 legally, can hold up the publication. I'm the one that  
23 has to then make a determination that, okay, we are not  
24 going to wait anymore, so we have a deadline of July 1

1 to return the final version of the paper to the Journal,  
2 and so that's the drop dead deadline for all of their  
3 discussions and review, no matter what, in my mind.

4 MS. MOORE: So July 1 it would  
5 actually be available?

6 DR. KEELER: That should be -- that's  
7 correct. It should be available around the July 1 time  
8 frame.

9 MS. MOORE: It would be pretty clear  
10 to you then if there were some other reason they didn't  
11 want to release it, not that that kind of thing would  
12 ever happen.

13 MS. BASSI: Certainly not in Illinois.

14 DR. KEELER: I think that's correct.

15 MR. ZABEL CONTINUES:

16 Q. Is my understanding correct, Doctor,  
17 that's the version you would be willing to release, the  
18 July 1 version?

19 A. The version that we will submit as a final  
20 version to the Journal will be the one I'm happy to  
21 share.

22 Q. Then I guess my only inquiry would be  
23 whether we can have Dr. Keeler back after our experts  
24 have had a chance to study that.

1           A.     It all depends on when.

2                   MR. RIESER:  It would be my  
3 suggestion, frankly, that, if it's going to be released  
4 by July 1, it would be my suggestion we just halt the  
5 cross-examination here --

6                   MADAM HEARING OFFICER:  We're not  
7 going to halt the cross-examination because, if it  
8 doesn't happen July 1, and we're right here arguing  
9 about whether or not the Steubenville study is going to  
10 be issued.  We're -- we'll continue on.

11                   Here's what I'm going to do.  I'm  
12 going to direct the Agency to provide a copy of the  
13 Steubenville report no later than July 5.  If it can't  
14 be available by July 5, then I will be willing to hear  
15 motions at that point.

16                   MR. ZABEL:  I will have a motion at  
17 that point, Madam Hearing Officer.

18                   MADAM HEARING OFFICER:  I just want to  
19 put it off, in case we can get to --

20                   MR. ZABEL:  I will reserve my motion,  
21 until the end of the Fourth of July holiday.

22                   DR. KEELER:  Can I just make one  
23 point?  I'm not going to be in town -- I'm actually going  
24 to be in the field, until about July 10, so I'm not



1 going to be dealing with this at all. I told you July  
2 1, just so that you would know that that's when we have  
3 to do it, but I'm not going to be in town to be able to  
4 do anything in regards to this, until I get back on  
5 about July 10. I think I get back to town late on July  
6 9, so putting a date of July 5 is not going to work for  
7 me. I'm just telling you. I have other commitments  
8 that I have made previously. This hearing got postponed  
9 from May, to now, and I had to cancel two different  
10 trips that I had planned for June in order to be here  
11 this week, and I can't cancel any more in the future  
12 because it involves field work, and I have contracts  
13 that are obligating me to be in the field.

14 MR. RIESER: We're just asking for the  
15 report, not you.

16 DR. KEELER: But if EPA responds back  
17 to me with whatever the final comments and all that kind  
18 of stuff very late in the game -- I'm hoping I get it  
19 before that, so I can get this taken care of, but if  
20 they get it back to me June 30 or something like that,  
21 I'm going to have a really hard time in dealing with  
22 that. I don't know what will happen. That's out of my  
23 hands.

24 MR. ZABEL: Madam Hearing Officer, I

1 will make my motion, regardless. I understand his  
2 difficulty. I'm giving up most of Father's Day with my  
3 children because of this hearing on Monday, so I'm  
4 sympathetic doctor, but at some point, we need the  
5 evidence in the record, and if it isn't there, I will  
6 make my motion.

7 DR. GIRARD: One problem we have got  
8 is that we have already got partial -- we have partial  
9 information in the record now, which you already said is  
10 not final. It seems to me that really need to talk with  
11 your attorneys to see if it is possible to put a draft  
12 copy, as it is now, in the record. There's no way it's  
13 going to be released for publication. We take trade  
14 secrets all the time in cases, so you need to talk with  
15 your attorneys. I think it would short circuit some of  
16 this questions by three or four hours, if you could  
17 simply look at what methods were used, and a lot of  
18 these questions go to procedures. They don't go to  
19 results at all. So I think you need to go talk with  
20 your attorneys.

21 (At which point, a break was taken.)

22 MADAM HEARING OFFICER: Back on the  
23 record.

24 MR. MATOESIAN: In response to the

1 question of -- on the question of Dr. Keeler's report,  
2 we're going to have to talk to the U.S. EPA about  
3 whether it can be released and what portion, so we  
4 cannot today commit to the July 5 date. We are going to  
5 reach out to them, and hopefully, get an answer by early  
6 next week, Monday, perhaps Tuesday. Then give you a  
7 more definitive answer then. That's the best we can do  
8 at this point.

9 MADAM HEARING OFFICER: Mr. Zabel.

10 MR. ZABEL: May I add to my request  
11 then that we also, if we receive anything, receive the  
12 200 pages or thereabouts of commentary that Dr. Keeler  
13 indicated he received from U.S. EPA? You are going to  
14 be talking to U.S. EPA, so you might as well talk to  
15 them about both.

16 MR. MATOESIAN: We will bring that  
17 issue up, as well, to them.

18 MR. RIESER: One other item, which I  
19 asked for yesterday, was if the LADCO presentation could  
20 be provided and I think Dr. Keeler said he would look  
21 for it.

22 DR. KEELER: It's being photocopied as  
23 we speak I believe.

24 MR. RIESER: Great. Thank you.

1                                   MADAM HEARING OFFICER: At this point,  
2 then, we will move on with the cross-examination of  
3 Dr. Keeler, and we will address issues considering the  
4 availability or unavailability of the Steubenville study  
5 published report at a later date.

6                                   MR. RIESER: I think we are on 12-A.

7                                   DR. KEELER: 12-A: "In your  
8 testimony, how do you define "local" and "regional"  
9 sources?" I believe I answered that previously. We had  
10 an extended discussion about that yesterday, so that's  
11 been answered. B: "Are the sources described in this  
12 state limited to coal-burning sources?" No, they are  
13 not.

14                                   DR. RIESER CONTINUES:

15                                   Q. What other sources are?

16                                   A. I believe I answered this yesterday, as  
17 well. Iron and steel production, municipal waste  
18 incineration, refining, metals production I think were  
19 the sources that I mentioned yesterday. C: "Is the  
20 deposition" --

21                                   Q. I'm sorry --

22                                   A. "Limited to wet deposition?"

23                                   Q. I'm sorry, one question on that. In  
24 identifying where the sources -- were the sources you

1 identified as a result of the receptor sampling and  
2 modeling or as a result of the review of the emissions  
3 inventory?

4 A. Only the receptor modeling.

5 Q. Thank you.

6 DR. KEELER: C, "Is the deposition  
7 described here limited to wet deposition?" The source  
8 apportionment results receptor modeling performed were  
9 only including the wet deposition?

10 MR. HARLEY CONTINUES:

11 Q. Dr. Keeler, mindful of the fact that the  
12 source apportionment was based on wet deposition, on  
13 days when it didn't rain, when there was only dry  
14 deposition, do you have an opinion about how that might  
15 have effected the results of the work that you did at  
16 Steubenville?

17 A. As part of this overall project, we will  
18 be looking at, both, wet and dry deposition. Again,  
19 this was the first paper that we attempted to get into  
20 the scientific journal, the wet deposition took  
21 precedence. The dry deposition work is ongoing, and  
22 it's going to be based on the extensive on-site  
23 meteorological measurements, as well as the reactive  
24 gaseous mercury, particulate mercury, and elemental

1 mercury measurements that we're actually making on site.  
2 We have, at this time, not modelled the deposition using  
3 the ambient data. However, based on our past projects,  
4 we would assume -- not one would assume. The dry  
5 deposition component should be a significant additional  
6 loading of mercury to this ecosystem. In our studies in  
7 Detroit, where we were in areas with similar source  
8 contributions, or similar source densities, the portions  
9 of dry deposition was -- well, the dry deposition that  
10 was about half of the overall deposition that we saw at  
11 the site at Steubenville was different in terms of the  
12 mix of sources than Detroit is, so we expect to see some  
13 different contributions, and so forth, but dry  
14 deposition is half the story we believe, and in an area  
15 with lots of sources, dry deposition could dominate the  
16 local deposition to this area.

17 Q. To be clear, that would be added to what  
18 you have already presented in your testimony about  
19 deposition through precipitation?

20 A. Yeah. The wet deposition -- just to be  
21 clear, the wet deposition that we modelled and think  
22 that 70 percent of which is coming from coal-fired  
23 utilities, is only about half of the total deposition,  
24 so mercury enters the waterbody by wet deposition, and



1 significant contributor to the dry deposition. I  
2 cannot, because I have not done that analysis, give you  
3 a quantitative number what percentage of dry deposition  
4 to coal-fired power plants is.

5 Q. Do things such as steel mills, metal  
6 working refineries have potentially and proportionate  
7 greater impact on dry deposition?

8 A. We're talking about dry deposition of  
9 mercury, right?

10 Q. Dry deposition of mercury in Steubenville.  
11 Keep it within the context of --

12 A. Again, local sources, if they are closer  
13 to the site and have different forms of mercury that  
14 they are emitting, would have a different contribution  
15 than coal-fired utilities would, yes.

16 Q. Thank you.

17 MR. RIESER CONTINUES:

18 Q. And during the 2003-2004 sampling, did you  
19 measure dry deposition in Steubenville?

20 A. We did not measure continuously for the  
21 2003-2004 time period a direct measurement of dry  
22 deposition that's analogous to what we did for wet  
23 deposition. What we have done and what we plan to do to  
24 provide that estimate is to use ambient concentrations



1 of mercury that have been measured, so we have  
2 continuous measurements of reactive gaseous mercury,  
3 elemental mercury, particulate mercury that are  
4 performed on an hourly time basis together with the  
5 onsite meteorological data, and we will model the dry  
6 deposition quantity to the surface at the Steubenville  
7 site. As part of intensive periods, we have direct  
8 measurements of mercury of dry deposition using  
9 surrogate surfaces, and other techniques, which all of  
10 these techniques have uncertainties, so we compare those  
11 measurements together with models to come up with a best  
12 estimate for dry deposition of mercury in Steubenville,  
13 but again, that analysis is not completed.

14 MR. HARLEY CONTINUES:

15 Q. In talking about the conduct of your  
16 ongoing research, you testified yesterday that you have  
17 assembled, but not compiled data for wet deposition for  
18 the year 2005. Is that correct?

19 A. The database is not finalized. We haven't  
20 compiled all of the trace element data and mercury data  
21 and ion data at this point.

22 Q. Based on your firsthand review of the data  
23 as it now exists, do you have an opinion as to whether  
24 or not the information from 2005 is consistent with what

1           you saw in 2004, 2003, or is it just too early?

2           A.       We haven't done any source apportionment  
3 modeling on the 2005 data at all because the data is not  
4 completed. The total amount of mercury deposition in  
5 Steubenville in 2005 which is the data that is done I  
6 don't have the exact number in front of me, but it's  
7 greater than the average. It's more than 2003, and I  
8 think less than 2004, so it's in the same vicinity.

9                               MR. BONEBRAKE CONTINUES:

10           Q.       Is the relative proportion of different  
11 species of mercury deposited different in wet  
12 deposition, as opposed to dry deposition?

13           A.       The answer to that question -- I'm going  
14 to try to simplify it. The reactive gaseous mercury  
15 forms and the particulate mercury forms are the ones  
16 that will most readily go into precipitation. When we  
17 analyze our samples for our precipitation samples for  
18 these species we see a very wide range in how much  
19 reactive particulate is there and part of that is  
20 because the reactions occur in the precipitation sample,  
21 itself, giving us somewhat an uncertain answer as to how  
22 much particulate was there, in the first place.  
23 Reactive mercury is also going to dominate the dry  
24 deposition because of its properties. It has a very

1 high deposition velocity, and so similar to wet  
2 deposition, reactive mercury is going to dominate in wet  
3 deposition, and also going to dominate in dry  
4 deposition, so in the absence of precipitation, you  
5 would expect to see a fairly rapid depreciation of the  
6 reactive mercury to the surface similarly as if it was  
7 going into cloud water, so I would expect reactive  
8 mercury to dominate, both, the wet and the dry. The one  
9 part, the dry deposition, that is less well understood,  
10 and is not included in the models, and the models that I  
11 have reviewed in the current literature is the uptake of  
12 elemental mercury by forest canopies and into the plant  
13 materials. This form of mercury and this deposition  
14 loss is something that's not characterized well, but  
15 it's a significant amount of mercury on an annual basis.  
16 It can be as much as three times the wet deposition to a  
17 forested ecosystem. It's not a simple answer.

18 Q. The deposition -- the mercury uptake in  
19 the forest canopy that you just mentioned, would you  
20 anticipate that that ultimately ends up in the  
21 environment, and if so, what portion of the environment,  
22 aside from the trees, themselves?

23 A. Well, what we've quantified thus far is  
24 the mercury that's taken up in deciduous leaves and

1 confer needles and so forth in the forest canopy. That  
2 material, at least, for the deciduous trees -- every  
3 fall, that material drops from the trees and falls to  
4 the forest floor, so mercury gets bound up in the  
5 organic matter and that material is bound to decompose  
6 like the other plant material and start working through  
7 the process of weathering and so forth, go into the  
8 soils, and some of it will run off, depending on the  
9 ecosystem structure, so it's a delayed signal, in terms  
10 of deposition to the forest canopy uptake through the  
11 growing season, deposition and then a slow process where  
12 it will then slowly make its way to the ecosystem.

13 MR. ZABEL CONTINUES:

14 Q. Just for my benefit, Doctor, you mentioned  
15 in discussing dry deposition the use of surrogate  
16 surfaces. Explain to me what that is or what that  
17 means.

18 A. Sure. One of difficulties in making  
19 measurements of the deposition of any pollutant, whether  
20 it's mercury or sulfur or any pollutant that's in the  
21 atmosphere, is that the surface of the earth is complex.  
22 Trees present a very complex surface in an urban area,  
23 buildings and roads and sidewalks, all those surfaces  
24 are very complex, in terms of the surface area, the

1 composition, how they move, how the wind and meteorology  
2 influences those, and so we can't make a replicate of a  
3 building and stick it up and then somehow wash the  
4 building down to get how much pollutant was deposited  
5 into that building, so we come up with a surrogate,  
6 something what we hope will mimic the surface, so we use  
7 a variety of different surfaces, and when we do this, we  
8 try to provide surfaces that have properties that won't  
9 artificially enhance the deposition, so there are other  
10 dynamic surfaces. Sometimes we will use a water surface  
11 if we are trying to look at and estimate the amount of  
12 mercury that's deposited into the Great Lakes, for  
13 example. If we're trying to assimilate the deposition  
14 to a specific type of surface, we would put that  
15 material on a small aerodynamic surface and actually  
16 measure how much deposits to that, and then to verify  
17 that we're getting reasonable results, we actually then  
18 use models to estimate how much mercury deposited to  
19 that small surface, and then use an understanding of the  
20 landscape and some of the information that we can get  
21 from a remote sense to tell us about how the landscapes  
22 change and the proportion of different types of trees  
23 and vegetation and plants and so forth to come up with a  
24 larger estimate of the deposition.

1           Q.     So in doing that, you use multiple  
2           surrogates, I take it, for different types of surfaces?

3           A.     We use more than one type of surrogate  
4           surface, that's correct.

5                     MADAM HEARING OFFICER:  Anything else?  
6           Moving on to D, I believe.

7                     DR. KEELER:  D:  "What meteorological  
8           analysis was performed to demonstrate this statement?"  
9           We used a combination of meteorological analysis tools  
10          to perform the meteorological interpretation for this  
11          project.  This included detailed plotting analysis of  
12          surface and upper air meteorological maps, backward air  
13          mass trajectories, NEX-RAD precipitation data sets,  
14          together with statistical techniques, such as cluster  
15          analysis to institute meteorological influences on the  
16          deposition.

17                    MR. RIESER CONTINUES:

18           Q.     If you look at Exhibit 29, which is the  
19           Beijing, China, Landis Report, there are -- there's a  
20           page after the page we looked at before called "Hy-split  
21           Back Trajectories."

22           A.     Yes.  That's what I was referring to in my  
23           statement.

24           Q.     So these are two examples of the back

1 trajectory?

2 MADAM HEARING OFFICER: Point of  
3 clarification, the Beijing study is Exhibit 30. 29 is  
4 the preliminary results. That's right. 30 is the  
5 Beijing study.

6 MR. RIESER CONTINUES:

7 Q. I had two identical exhibits. Sorry about  
8 that. Exhibit 30. Doctor, you're there. "Hy-split  
9 Back Trajectories."

10 A. Yes.

11 Q. So these are examples of the type of work  
12 that you would do?

13 A. These are examples of one of the tools  
14 that I listed there. These are the -- as I said, the  
15 backward air mass trajectories.

16 Q. What information do using these as  
17 examples -- what information do these provide to you?

18 A. If you look at these plots, these plots,  
19 basically, the trajectories are calculated to correspond  
20 to the maximum hour of the precipitation for a couple of  
21 the precipitation results we received in Steubenville,  
22 and these black lines represent the center of  
23 probability of mass that air would have had thought  
24 going backwards from Steubenville and where the air mass

1 would have come from moving backwards in time and I  
2 think it's six-hour increments, and that's what those  
3 little stars are.

4 Q. The stars are six-hour increments?

5 A. I believe so, yes. That's going back 72  
6 hours. These are three-day back trajectories.

7 Q. Then there are graphs below these that are  
8 on the side that says "HPA." Would you tell us what  
9 these rep?

10 A. Yes. The bottom plots HPA is a unit of  
11 pressure, and so this gives us a vertical slice looking  
12 up in the atmosphere of the height that the air mass was  
13 predicted to be at, so again, if you start at  
14 Steubenville, or to the right, looking at the first here  
15 moving backwards in time, it shows that our site was at  
16 the surface, and then the air mass actually came moving  
17 backwards in time, and went up into the atmosphere a  
18 little bit into the higher into the mix layer and then  
19 up as it went backwards and slowly came down again.

20 Q. Did these examples -- again, do these also  
21 represent the difficulty in identifying the proximity of  
22 regional source because the three days' time travelled  
23 put you pretty far away from Steubenville? Can you be  
24 more specific about the difficulties you had to -- why



1 do you select three days? Let me ask that.

2 A. For the purposes of this demonstration,  
3 three days was selected because, when you start going  
4 back further, you actually have missing data in the data  
5 streams, and you wind up actually having the model  
6 condition calculate very often, so we use three days  
7 because, as part of that the original analysis we did  
8 back in the late 80's looking at some of the utility  
9 acid precipitation data, we found that three days  
10 represented some synoptic meteorological conditions very  
11 well, and that going further -- actually, the  
12 uncertainty in the trajectory calculation was back much  
13 more than three days which was so great to make them  
14 unreliable.

15 Q. Looking up, again, at the specific  
16 question of 12, "Meteorologic analysis corroborates that  
17 a substantial amount of the mercury deposition found at  
18 the Steubenville site was due to local and regional  
19 sources," would the local and regional sources be  
20 sources that contribute mercury emissions to the  
21 atmosphere or all along the lines of these back  
22 trajectories?

23 A. The line here, again, is meant to  
24 represent the most probable path, so as you move

1 backwards in time, the path that actually the air mass  
2 could have taken actually grows wider, so if you were to  
3 take and draw an increasing, like, wedge of air moving  
4 backwards in time, it actually encompasses a larger area  
5 than what's shown by just this line. This is just a  
6 representation, and so what we do then is to use this as  
7 a guide, and then take the surface and upper air  
8 meteorological maps together with the NEX-RAD to help  
9 this in terms of guiding us in terms of the timing for  
10 how quickly the air mass has moved, where the  
11 precipitation fell along this trajectory, and what  
12 sources could have fallen into this kind of large area  
13 that surrounds this line that you see here, and so you  
14 put that all together and together with the known source  
15 locations to be able to then say, "Here's the sphere  
16 that we think has contributed to the deposition at that  
17 point."

18 Q. When you -- tell us what "NEX-RAD" is, I'm  
19 sorry.

20 A. NEX-RAD is the weather radar that you see  
21 presented on the television. It's the nice way to show  
22 the storms for Springfield, show the nice, heavy  
23 thunderstorms that come through in different colors, so  
24 it's next-generation radar is what it stands for.

1           Q.     So going back to your description of this  
2     you describe sort of a -- since I'm used to ground water  
3     what I will call an inverse plume where it spreads out  
4     from the back and focuses in at the front end. Is that  
5     an apt description?

6           A.     I'm not a ground water person, but plume  
7     is okay.

8           Q.     So back to that question, would it be in  
9     the way you did the study that all of the sources that  
10    you identify within this area, looking at this first map  
11    on the left, which starts at, essentially, the Rio  
12    Grand, all the sources within that area leading up to  
13    Steubenville would be included in your analysis of the  
14    mercury emissions that fall at Steubenville?

15          A.     We will utilize going as far back as the  
16    trajectory would tell us for the three days. We  
17    actually look at the meteorological numbers for the  
18    entire country when we are doing this, but we would look  
19    at and investigate any potential influences that would  
20    occur along that entire path.

21          Q.     So when you talk about local -- we had an  
22    extended discussion yesterday I understand about local  
23    and regional sources and things like that, and you were  
24    -- again, correct me if I'm mistaken -- but you were

1 concerned that the local and regional as the proximity  
2 issue was problematic because, in your mind, it really  
3 wasn't meteorologic?

4 A. That's correct.

5 Q. The meteorology that's described in this  
6 first plot, the one on the left on this Exhibit 30,  
7 would suggest that the mercury that's being contributed  
8 into Steubenville that you measured came from sources  
9 along this line and then spread out as far as the data  
10 shows that the air mass moved. Is that correct?

11 A. I believe what you're trying to convey is  
12 correct.

13 Q. Is it, again, looking at this map, is it  
14 the assumption that you're going back three days because  
15 that's an understanding of what's useful data? Is that  
16 the way to say that? After three days it gets mushy?

17 A. Yes.

18 Q. It doesn't really tell you anything. Do  
19 you have any assumption, and I asked this yesterday, but  
20 there is a good graphical representation of it. Do you  
21 have any assumptions that any amount of mercury comes  
22 into this back end that you have got at the Rio Grand in  
23 looking at how much mercury is being deposited in  
24 Steubenville?

1           A.     Again, let me see if I understand what  
2     you're asking me.  This line, or whatever, that goes  
3     back toward the Rio Grand here, we don't add up  
4     emissions along that line or suggest that emissions  
5     are -- that's not included in our analysis.  There are  
6     no assumptions in terms of the analysis.  What we would  
7     do is we would take and look at the entire upwind  
8     history of this precipitation event.  One tool to look  
9     at where the air mass came from would be this line, and  
10    along this line we would look at things, such as did it  
11    rain?  What was the precipitation rate moving backwards  
12    i time along this line, and what we found is that, by  
13    using precipitation, you can look to see whether the  
14    probability that the pollutant would be removed is  
15    greater in times when precipitation along these back  
16    trajectories is high or not, so we look at the storms,  
17    and in this case, this one that's for May 8, it turned  
18    out that the air mass actually wasn't moving very fast.  
19    If you look at those stars there, there are a few stars  
20    very close to Steubenville, so the air mass was not  
21    moving very fast.  In fact, those first four or five  
22    stars there reflect the meteorological situation at the  
23    time, which basically, had a storm which stalled out in  
24    that area, so the storm was a big storm.  Things didn't

1        move very quickly and the precipitation would have been  
2        washing out all of the reactive mercury, and all of the  
3        other very reactive pollutants from the atmosphere and  
4        cleaning things up for a very long time, so that the air  
5        masses that could have contributed to the loading that  
6        was occurring through that storm couldn't come from very  
7        far, and so this is how we do that type of analysis, and  
8        one can take and model that specific storm and compare  
9        that against using observations because there are no  
10       assumptions in the observations. It's meteorological  
11       data that we have and we don't look at a line underneath  
12       the map. We actually look at the whole area, and we,  
13       basically, say, from this, we know that we had a very  
14       slow-moving air mass that took a long time for it to  
15       actually go anywhere, so we are talking about emissions  
16       that were in the vicinity of the Steubenville area. The  
17       ones that were coming from Southern Ohio that were --  
18       the emissions sources were, basically, emitting up into  
19       that area, and then that storm was staying there, and it  
20       was precipitating out for an extended period of time, so  
21       that the air mass, by the time it got up towards,  
22       Steubenville had already been washed out. The air that  
23       was in that storm had already had a great deal of  
24       removal and cleansing, just to put it in simple terms,

1       so the highest probability of where the air masses that  
2       fed the clouds could have come from had to have been in  
3       that general vicinity.  Again, I'm saying Southern Ohio,  
4       Northern Kentucky, that general area, based on the  
5       service meteorological data our on site, data as well as  
6       using this as a tool.  When you start talking about  
7       meteorological analysis and showing maps and so forth,  
8       everyone turns off and goes to sleep in your talk, but  
9       when you show a trajectory plot, people get the idea  
10      that the air started down somewhere in Texas and that's  
11      the path that it took before it got to Ohio, so it's  
12      simplification for presentation purposes, but by no  
13      means is it what we use as the definitive answer because  
14      there's very complex three-dimensional flows that occur  
15      in the atmosphere, and one has to take those into  
16      account, as well, when one is looking at what potential  
17      source areas could have contributed.

18                Q.     So that description is what tells you that  
19      the mercury deposited in Steubenville came from the  
20      specific area you described, Southern Ohio, Kentucky,  
21      and not from, say, Houston, which is more or less passed  
22      -- is that correct?

23                A.     Yes, in simple terms.

24                Q.     In doing that analysis, do you have to

1 take into account the atmospheric chemistry in  
2 transformations of mercury in the atmosphere?

3 A. We do rely on our understanding of the  
4 reactivity of mercury and the chemistry of mercury in  
5 order to be able to look at what distance scales we  
6 think things could take place on. However, there are no  
7 assumptions made. We use what our best available  
8 understanding is based on, both, our observation data  
9 that we have taken, both, at Steubenville and at our  
10 Michigan sites, together with what is published in the  
11 literature.

12 Q. And is that understanding -- you use the  
13 term "understanding." Is that correct?

14 A. I believe that's what I said, yes.

15 Q. Is that understanding different than the  
16 atmospheric transformation of mercury which are  
17 described in the deterministic models?

18 A. That would depend upon the specific  
19 deterministic model. This gets to the whole question  
20 that I raised yesterday about the uncertainties in the  
21 models.

22 Q. Let's use CMAQ to narrow it down?

23 A. CMAQ -- the version of CMAQ that EPA uses  
24 has a very simplistic atmospheric chemistry for mercury.



1       There are lots of issues in that chemistry. It does not  
2       include all the relevant reactions that we understand  
3       today. It doesn't include the most up-to-date reaction  
4       rates, so I would say no. We rely upon more than that.

5               Q.     And have you published or is there another  
6       paper published paper that you rely on that identifies  
7       the atmospheric transformation reactions that you rely  
8       on in doing your work?

9               A.     We have papers. We have a paper that we  
10       have submitted to -- see, we have a paper that we  
11       submitted to a peer-reviewed scientific journal that  
12       details some of our deterministic modeling, which is not  
13       at all what was done here, but that's not really  
14       relevant to the situation, but it incorporates some of  
15       the mercury chemistry that we rely on. I mean, some of  
16       these chemical reactions and updating the atmospheric  
17       chemistry happened very rapidly. I was just in Russia  
18       at a mercury meeting a long-range transport mercury  
19       meeting in Moscow and learned of some new reactions that  
20       had just come out, so we rely upon, both, what our  
21       colleagues tell us at meetings, as well as what the  
22       peer-reviewed literature has. I don't write down, and I  
23       don't have, like, a publication where I have listed all  
24       the atmospheric chemistry that I rely on. I rely on

1 what's in the peer-reviewed literature.

2 Q. I guess what the bottom line of what I'm  
3 trying to get at is what Sheldon would say is, is there  
4 a way that we can see how you describe the atmospheric  
5 transformation reactions? I assume there is some type  
6 of mathematical equation or part of your model or  
7 something that another person can look at and apply in a  
8 similar setting and test.

9 A. We don't use chemical reactions in our  
10 modeling. I stated yesterday that the receptor modeling  
11 does not use chemical reactions. We only used observed  
12 information.

13 Q. Then I completely misunderstood because I  
14 thought you said that, in evaluating where mercury came  
15 from, you made certain judgments, decisions about the  
16 atmospheric chemistry that mercury undergoes.

17 A. What we use as our understanding that  
18 reactive mercury is removed very easily with  
19 precipitation, which we have published and have  
20 observations that show this, and that elemental mercury  
21 is not removed as rapid, so again, this is based on  
22 observation of published work, and in understanding the  
23 information, we have an idea of how long reactive  
24 mercury would last in the atmosphere based on almost

1 instantaneous removal in precipitating systems. That's  
2 the atmosphere chemistry that I'm referring to is that  
3 there are different chemical properties of mercury forms  
4 that may get removed at different time scales.

5 Q. Do you have a numeric rate that you use to  
6 evaluate the removal of reactive gaseous mercury?

7 A. No, sir.

8 Q. When you say it's rapidly removed, what's  
9 the quantification of rapidly?

10 A. Well, we have hourly data, and you will  
11 see within an hour time frame a very rapid, so within an  
12 hour time frame, we see a significant -- more than 50  
13 percent -- removal of the mercury within one hour of the  
14 onset of precipitation.

15 Q. When you talking about that data, what  
16 you're talking about is the data at your wet deposition  
17 sampling locations, correct?

18 A. That's correct.

19 Q. So the storm begins at five o'clock. At  
20 5:15, you have a sample that's got a lot of mercury in  
21 it. At 6:15, there's no mercury, just --

22 A. The reactive mercury would have dropped,  
23 that's right.

24 Q. And what does that -- but is it correct

1           that you use that data to make decisions as to how much  
2           reactive gaseous mercury is in the air mass that's  
3           moving into Steubenville?

4           A.     No.  There's no assumption there.

5           Q.     So how do you know how much mercury is in  
6           the air mass that's moving into Steubenville or where  
7           it's from?

8           A.     We have on-site measurements of the  
9           reactive mercury at the site.  There's no assumption.  
10          That is my point.

11          Q.     What does it tell you -- what does that  
12          data tell you about -- well, step back.  That's why  
13          you're not in position to identify the proximity of the  
14          sources based on the samples that you do at  
15          Steubenville, correct?

16          A.     We have not done a detailed source  
17          apportionment of the ambient mercury at Steubenville,  
18          yet.  What we have done at this point is look to see  
19          specific episodes when we have high reactive and  
20          particulate mercury to see what the meteorological  
21          conditions were like and where the air was coming from  
22          based on on-site meteorological measurements, as well as  
23          other meteorological data, such as the trajectories, to  
24          see if there is a strong association, so whether we see

1 repeated pattern of things like high sulfur dioxide,  
2 high reactive mercury, high nitrous oxide, high CO,  
3 whatever the on-site measurements we have, high  
4 particulates, and then we analyze the particulates for  
5 there elemental composition, as well, so we can look for  
6 the same tracers in the air as we see in the  
7 precipitation, and that analysis has not been completed,  
8 yet, because we collect four samples a day. That would  
9 collect them for -- well, we're on our third year now,  
10 and it's thousands of samples that we are in the process  
11 of analyzing, so that analysis and information will  
12 come. We do see strong relationships that winds that  
13 flow from the south, southwest with higher mercury  
14 levels and higher SO<sub>2</sub> (phonetic) concentrations that are  
15 indicative of transport from fossil-fuel-burning plants  
16 from coal-burning plants.

17 Q. What tells you that those are fossil  
18 burning plants, again, looking at this first map as an  
19 example in Kentucky, Ohio River Valley and not Houston?

20 A. Well, we have an idea of the location of  
21 where the fossil-fuel-burning facilities are, where all  
22 the sources are, and then we can look along that swath  
23 that we think where the air could have possibly come  
24 from, and we look at the inventory to tell us what the

1 possible sources could have been. It's an aid, not as a  
2 quantitative tool.

3 Q. Then you have to make -- then you have to  
4 evaluate whether precipitation events have washed out  
5 mercury that could come from other sources along that  
6 line. Is that correct?

7 A. That's right. It's something I have been  
8 working on for more than 20 years, so in my best expert  
9 judgment, I have done this for multiple pollutants,  
10 including mercury, and I feel like I have a very good  
11 handle on this situation, and on this type of analysis.

12 Q. If there was an example where there was an  
13 air mass moving into Steubenville, and there was no rain  
14 fall events going on along the line back to Texas, what  
15 would the results of Steubenville look like?

16 A. You know, I would have to go and look to  
17 see if we had those type of events in order for me to be  
18 able to predict because you can draw the same line  
19 moving from Steubenville to Texas, and depending upon  
20 the winds, how strong the winds were, and at all those  
21 segments, what the atmosphere was like, the stability of  
22 the atmosphere along the trajectory, the temperature  
23 profile, all these different meteorological parameters  
24 would affect disbursement, the chemistry, the transport,

1 and the dry deposition that occurred along that  
2 trajectory. One of the things that, if one was to take  
3 and estimate how much mercury you would expect left in  
4 an air mass that was submitted to Texas, by the time it  
5 got to Steubenville, one can take and estimate how much  
6 you would have expected to have loss from dry  
7 deposition, how much you might expected to have lost  
8 from chemical reactions, and just by the air mass being  
9 deluded as you got there and when you get that far back  
10 in time, there's very small contribution that you would  
11 see in Steubenville from sources way back in Texas, but  
12 that's a general comment. I would have to look at  
13 specific information to give you a quantitative answer.

14 Q. In doing that estimate that you just  
15 described, one of the factors would be the atmosphere  
16 transformation that mercury undergoes --

17 A. Yes.

18 Q. -- during that time of travel?

19 A. Yes, sir.

20 MR. HARLEY CONTINUES:

21 Q. The May 8 event that you described on your  
22 hy-split trajectory, is that one of the days during that  
23 year where there was an event that precipitation led to  
24 a significant deposition of mercury in Steubenville?

1           A.     Yeah.  That was one of the four largest  
2           events depositing.  Again, I think I've given a figure  
3           of maybe the four largest or the several largest, as  
4           much as 8 percent of the annual total, so these are  
5           significant events that occurred at this site.

6           Q.     In the two-year period of 2003, 2004, how  
7           many total mercury rains have there been on Steubenville  
8           comparable to the event on May 8 where we have  
9           hyperloading of mercury on Steubenville?

10          A.     Gee, I don't have that in front of me.  My  
11          memory is that we had in the two-year period about eight  
12          events that were greater than .6 micrograms per meter  
13          squared I believe, which that's a big event.  And then a  
14          couple that were greater than one, and again, if you are  
15          looking at 10 to 20 micrograms per square meter at a  
16          year at a site, if you get more than one in one rain  
17          event, that's a significant contribution in that one  
18          day.

19          Q.     There was one other question I had.  The  
20          CMAQ model, you gave some reasons why the CMAQ model is  
21          it not as precise as the receptor-based approach that  
22          you used in Steubenville.  One question I had about the  
23          CMAQ model is you said the CMAQ model underestimated  
24          mercury loading in Steubenville by comparison to the



1 results of the work that you have done. Is that  
2 correct?

3 A. Yeah. Just to qualify, the CMAQ for 2001  
4 is where they calculated that 43 percent was coming from  
5 coal-fired utility boilers. Our data is from 2003,  
6 2004, we did have the opportunity to compare our  
7 Michigan network sites for the 2001 year, together with  
8 our estimated for Vermont, and the CMAQ model  
9 underestimated the deposition that we measured at our  
10 Michigan sites and at the Vermont site for 2001. It was  
11 off by a factor of two at one site, as I recall, so it  
12 grossly underestimated the total deposition at our sites  
13 that we had measuring in 2001. The way you phrased it,  
14 "CMAQ being less precise" I think the way I would rather  
15 phrase it is that these type of models are more  
16 uncertain, that there are more parameters and processes  
17 that are poorly defined in these models, which make  
18 their uncertainties greater and the conclusions drawn  
19 from those models much greater than I would say that  
20 they are from the receptor modeling. That's a correct  
21 statement.

22 Q. Is the CMAQ model a model which accounts  
23 for, both, wet and dry deposition of mercury?

24 A. Yes, it does.

1           Q.     And is it fair to characterize your  
2           testimony as indicating that, in the Steubenville  
3           situation, your modeling of wet deposition alone was  
4           greater than the CMAQ model of, both, wet and dry  
5           deposition at that site?

6           A.     My memory tells me that CMAQ actually  
7           estimated a greater dry deposition at Steubenville than  
8           it did for wet deposition for the 2001 year, so no, our  
9           wet deposition wasn't greater than the CMAQ's total. I  
10          believe the CMAQ total was around 30 micrograms per  
11          square meter, or 34. It's hard to tell from the  
12          modeling because they do it in colors with ranges, but  
13          that, again, I'm just going from my memory of what Russ  
14          Bullock presented, but I believe that that's true, that  
15          the dry deposition was greater in this area than the wet  
16          deposition was. Using the CMAQ model, not from our  
17          observations.

18                                   MR. BONEBRAKE CONTINUES:

19          Q.     On the May 8 back trajectory, a line, is  
20          that, essentially, indicative of a wind direction from  
21          the southwest?

22          A.     Yeah. The line that starts at  
23          Steubenville or the line that starts at the site there  
24          on the border of Pennsylvania and Ohio, if you follow

1 that backwards towards the bottom of the left page that  
2 is trying to illustrate the most likely path that an air  
3 mass took before it got to Steubenville.

4 Q. In this particular case, through Texas and  
5 Louisiana, the air mass was moving from the southwest  
6 and northeast?

7 A. That's correct, southwest.

8 Q. Yesterday we talked a little about wind  
9 directions in Illinois. My understanding from your  
10 testimony was that, in the winter season in Illinois,  
11 the wind direction is frequently from the northwest. Is  
12 that correct?

13 A. Yeah. The Great Lakes region, as a whole,  
14 has a higher frequency of winds from the north,  
15 northwest during the wintertime as a result of synoptic  
16 meteorological conditions.

17 Q. And during the summer months, are winds in  
18 the state of Illinois most typically from the southwest?

19 A. From the data I looked at for O'Hare, it  
20 appears as if the south is the dominant winds with  
21 south, southwest being one of the more frequent wind  
22 directions, but knowing that westerly winds are also the  
23 dominant wind areas during the summer.

24 Q. Would O'Hare's -- would the direction of

1 wind at O'Hare be impacted by Lake Michigan?

2 A. Yes.

3 Q. So would the data from O'Hare, in your  
4 view, be representative of the rest of the state of  
5 Illinois?

6 A. No. If you look at wind measurements made  
7 in Southern Illinois, you see a slight variation in that  
8 overall pattern. You might have slightly more winds  
9 from the south than you would at O'Hare, so the winds  
10 will change slightly. The dominant wind direction still  
11 is going to be west with a southerly component being  
12 strongest.

13 Q. With respect to the northwest portion of  
14 the state of Illinois, would you expect, then, that  
15 during the winter months the predominant wind direction  
16 would be from the northwest and then the summer months  
17 it would be from the southwest?

18 A. South, southwest. From what I have seen  
19 for -- and again, I haven't looked at a climatological  
20 average, but that's about right.

21 Q. What would you view, Dr. Keeler, to be the  
22 states which contain upwind sources of mercury with  
23 respect to the state of Illinois?

24 A. Again, taking what you said in terms of

1 the dominant wind directions, you would have the states  
2 of Missouri, Iowa, and farther to the west would be  
3 upwind of Illinois, and then the states to the south,  
4 Kentucky, Tennessee, Louisiana, Alabama.

5 Q. Would Texas also be in that list?

6 A. Texas would be in that list, as well.

7 Q. One other question for you, Dr. Keeler.

8 You mentioned in response to a question from Mr. Harley  
9 that you had done, as I understand it, a comparison of  
10 CMAQ predictions to Michigan deposition data that you  
11 had available to you. Is that correct?

12 A. Yes.

13 Q. Is that comparison in a published article?

14 A. No, it's not.

15 Q. Is that comparison publicly available?

16 A. You know, is it in Dr. Landis' briefing?

17 MADAM HEARING OFFICER: Exhibit 30,  
18 Beijing Exhibit 30, yes.

19 DR. KEELER: It says, "Comparison of  
20 CMAQ Model Versus" --

21 Q. Back toward the end of the document.

22 MADAM HEARING OFFICER: Yeah,  
23 two-thirds back he said. "Comparison of CMAQ Model  
24 versus UMAQL Measured Mercury Wet Deposition Testimony."

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24

MR. BONEBRAKE CONTINUES:

Q. Is that correct, Dr. Keeler, that's the page you're on?

A. Yes, and refers to the left column to Dexter, Michigan.

Q. Helston, Michigan?

A. That's correct.

Q. Both, the CMAQ predictions and the deposition data that you were using in the comparison were both from 2001?

A. Yes, sir. These numbers were provided to Dr. Landis I believe by Russ Bullock of U.S. EPA, and these are the actual data that went into the CAMR modeling.

Q. Is there a discussion of this comparison in the study that's been at issue today in terms of what --

A. No. This is an additional analysis. The Michigan site data is not included in the Ohio paper.

Q. Other than what's in this particular page, has the comparison or the bases of the comparison otherwise been made publicly available?

A. All of the Michigan data for 2001 is in peer-reviewed publications and the CMAQ model results

1 are in the public docket that was filed by IEPA.

2 Q. So --

3 A. So it's all public.

4 Q. From your perspective, then, someone could  
5 take the publicly-available information in those  
6 documents and repeat the comparison?

7 A. Yes, sir.

8 MR. ZABEL CONTINUES:

9 Q. Earlier I think you said, in doing the  
10 corroboration using a Michigan inventory, you looked at  
11 sources east of the Mississippi. Is that correct?.

12 A. I said that was the area that we focused  
13 on, yes.

14 Q. You didn't use, for a particular  
15 corroboration, also sources east of the Mississippi?

16 A. Again, for trying to speak generally to be  
17 inclusive of all the work that we did for a particular  
18 event, we would use our understanding of the  
19 meteorologic and transport for a particular period, such  
20 as the May 8 period, where we would focus on those  
21 sources, and sources that were to the southwest, for  
22 example, in the most highest probable transport area, so  
23 we wouldn't be then taking out the list of sources from  
24 Minnesota and Wisconsin at that time to look at -- in

1 order to think about the May 8 event. On another day  
2 where the flow came from those directions, we would look  
3 at another set of sources that would help us interpret  
4 that specific day. We didn't limit our analysis to a  
5 number of sources. We had the entire -- plotted all of  
6 the sources plotted in 1999 EPA database together with a  
7 list of all those sources and the estimated fraction of  
8 mercury emitted from each source in each fraction as a  
9 tool to guide us in terms of our interpretation.

10 Q. That was sort of the point of my question,  
11 and the May 8 is a good example. There's some subset of  
12 sources east of the Mississippi you would have used in  
13 looking at emission inventories for them?

14 A. Again, we would have focused on the  
15 sources in the states which we thought had a probability  
16 of contributing on a specific day for a specific event.

17 Q. Just to make this perfectly clear to me,  
18 there are coal-fired plants in Tampa Bay in Florida.  
19 They would not have been included, would they, in your  
20 May 8 analysis?

21 A. I would say that I did not look at Tampa  
22 Bay power plants on interpretation of the May 8.

23 Q. On that May 8 one, both, Louisiana and  
24 Texas I believe are west of the Mississippi, were they



1 not?

2 A. That would be a geographically-correct  
3 answer.

4 Q. And in fact, Minnesota -- for instance, if  
5 you're looking in the other direction, although I  
6 understand Minnesota has a dispute on which side of the  
7 Mississippi they are in, at least, part of it is west of  
8 the Mississippi, is it not?

9 A. Yes, it is. In my use of east of the  
10 Mississippi and west of the Mississippi, I was trying to  
11 give a general division line, but as I mentioned  
12 earlier, I was not trying to distinguish an absolute  
13 line of demarcation for any of the analysis.

14 Q. Well, then in the May 8 case, would you  
15 have looked at cases in Louisiana and Texas?

16 A. We would have looked at sources along that  
17 trajectory, yes, or in that area that would be indicated  
18 by that air mass trajectory, yes.

19 Q. In answer to one of the questions I  
20 understand that cone gets wider the far away from  
21 Steubenville you get?

22 A. That's right. So sources all the way up  
23 from Kansas to Atlanta, Atlanta area, Georgia would also  
24 have been considered in looking in the backward upwind

1 path.

2 Q. Most of Texas, I take it, at that point?

3 A. Yes.

4 Q. Thank you.

5 MADAM HEARING OFFICER: Anything  
6 further? I believe 12-E --

7 MR. RIESER CONTINUES:

8 Q. This will actually address some of the  
9 questions, but since Mr. Harley brought up the  
10 comparison of CMAQ and the receptor modeling, it's  
11 correct, isn't it, that CMAQ in your type of receptors  
12 study are designed to achieve two different results. Is  
13 that correct? It's designed for two different purposes.  
14 Is that correct?

15 A. How a model is used is defined by the user  
16 and so in my case, which I can attest to, we were using  
17 receptor models to calculate how much mercury was  
18 deposited via wet deposition from the major source  
19 categories for the period of 2003-2004. So that  
20 apportionment to source categories was our objective,  
21 and that's where we did the modeling. Why EPA did the  
22 CMAQ modeling for 2001? As I understand, it was to  
23 estimate the contributions from all the major sources to  
24 the deposition of mercury across the entire United

1 States, so by definition, it has a broader purpose, and  
2 was not looking at one site. It was not looking at just  
3 wet deposition. It looked at wet, dry, and ambient, so  
4 yes, they have -- they have different purposes, but to  
5 say CMAQ model has one purpose and the type modeling we  
6 did had only one purpose, it really depends on what the  
7 user was using that model for.

8 Q. The CMAQ model, among its utilities, is  
9 that it can be used for predictions. Isn't that  
10 correct?

11 A. Yes, sir.

12 Q. So you can use it to decide what would  
13 happen, as an example reflected in Exhibit 30, "Utility  
14 zero out," of what mercury deposition would look like if  
15 there were no utility emissions?

16 A. Yeah. I believe, like you said, in  
17 Exhibit 30, I believe that map included a couple special  
18 plots. Maybe I'm wrong.

19 Q. You're correct. There is a CMAQ simulated  
20 total mercury deposition for 2001 and underneath it  
21 says, "Utility zero out," and prior to that, there's a  
22 base case CMAQ simulated total mercury for 2001 base  
23 case, so as we said, you can use CMAQ to, as an example,  
24 take out all of the utility emissions and see what

1 things look like?

2 A. Correct.

3 Q. Can you use your model to do that?

4 A. The receptor model, by definition, does  
5 not have a predictive capability.

6 Q. So if I -- well, let me ask you, if you  
7 assumed that the coal-fired power plants, within a 50  
8 kilometer range of Steubenville, ceased to operate,  
9 would you have any conclusion as to what that would do  
10 to the mercury deposition in Steubenville?

11 A. Based on our work, what I would say is  
12 that if the coal-fired utilities, regardless of  
13 location, stopped emitting, so they were zeroed out, we  
14 would see close to a 70 percent reduction in the mercury  
15 deposition that we measured at Steubenville.

16 Q. Could that conclusion, in making that  
17 statement, what you were referring to are not the  
18 coal-fired utilities within a 50 kilometer distance from  
19 Steubenville, but all coal-fired utilities in the United  
20 States?

21 A. We -- again, based on our analysis, our  
22 meteorological analysis, we will consider sources that  
23 were beyond regional, so very long range sources, so  
24 ones that are the western part of the United States

1 would contribute very little to that total and part of  
2 that is backed up by just looking at the deposition data  
3 that's in the Western United States, which is very low.  
4 It's for around four micrograms per square meter, so I  
5 would say that it's the sources in the eastern United  
6 States coal-fired utilities in the Eastern United States  
7 which are contributing that 70 percent, and if those  
8 were reduced, we would see that commensurate drop in the  
9 mercury deposition at that site for those years that we  
10 modelled.

11 Q. But you wouldn't be able to use your  
12 receptor model to determine what would happen at  
13 Steubenville if, for example, all the power plants in  
14 the state of Ohio reduced mercury emissions by 70  
15 percent?

16 A. No. The model is not capable of  
17 predicting calculations.

18 MADAM HEARING OFFICER: Are we ready,  
19 then -- I believe we have answered 12-E and I think  
20 perhaps 12-F already. In the discussion we had this  
21 morning, E is, "Does the analysis differentiate between  
22 sources located at different distances?" and we have had  
23 substantial discussion about --

24 MR. RIESER: Correct, yes.

1                   MADAM HEARING OFFICER: F is, "Have  
2 you quantified the substantial amount as used in this  
3 statement?"

4                   MR. RIESER: I was looking at the  
5 wrong number, yes, correct.

6                   MADAM HEARING OFFICER: And we  
7 answered 13-A and B yesterday, so 13-C.

8                   DR. KEELER: I actually believe I  
9 answered this one as well. "Are they fired with  
10 bituminous or sub-bituminous coal?" Again, in the  
11 complete listing of the emissions sources and the EPA  
12 inventory, it lists the type of coal burned. I believe  
13 there is a mix of plants that burn, both, bituminous and  
14 sub-bituminous. Some are all sub-bituminous and some  
15 are -- a few that are bituminous, so there's a  
16 combination. D: "Does the chemical profile of  
17 bituminous as opposed to sub-bituminous affect the  
18 amount and type of mercury emitted by a coal-burning  
19 generating unit?" Again, I believe I answered this  
20 yesterday in saying that I'm not an expert on control,  
21 and so perhaps, an expert for the State will provide --  
22 later will talk about how various control will affect  
23 the emissions, but clearly, the type of coal burned will  
24 have an influence on the type of type and form of

1 mercury, in respect to whether it's reactive mercury or  
2 gaseous mercury or particulate mercury and we talked  
3 about the importance of chlorine and fly ash content and  
4 iron and others things in the coal that would cause  
5 those differences. E: "Would you expect a different  
6 result at Steubenville if the surrounding units burned  
7 sub-bituminous coal?" I don't have the ability to  
8 provide an answer to that question. I don't understand  
9 exactly what you are asking. I assume that the power  
10 plants in the area are not -- from an inventory, some do  
11 burn sub-bituminous coal, so I'm not sure if you are  
12 asking me if all the plants in the Steubenville area  
13 burn sub-bituminous but I see --

14 Q. I am asking, if all the power plants  
15 surrounding Steubenville burn sub-bituminous coal, would  
16 you expect a different result?

17 A. Again, I guess I'm not prepared to answer  
18 that question. I think I would have to have more  
19 understanding of how changing to a different coal type  
20 for the various types of controls that are used in that  
21 area would affect speciation and I'm not prepared to  
22 answer that.

23 MADAM HEARING OFFICER: And I believe  
24 you have answered 13-F, as well, whether airborne

1 sources of mercury are located within 50 miles.

2 DR. KEELER: So G: "In what way, if  
3 any, are the conditions at Steubenville analogous to the  
4 conditions in Illinois?"

5 MR. RIESER: To be honest, there was  
6 some movement in that answer, so if we get too  
7 repetitive, cut us off, but I would like to go through  
8 this because I heard two different things, to be honest.

9 MR. KIM: Could you clarify what you  
10 mean because Dr. Keeler did describe the distinctions in  
11 terms of meteorological conditions and so forth and I  
12 recall him going into something about topography. Were  
13 you wanting something beyond that, something different?

14 MR. RIESER: What we're looking for is  
15 whether the findings with respect to Steubenville are  
16 analogous to Illinois, and some of that discussion has  
17 to do with things that were discussed and things weren't  
18 discussed. I heard, on the one side, Steubenville was  
19 typical, but in some ways, it's obviously atypical due  
20 to the amount of power plants that are around, so I  
21 think we need to walk through each of the features and  
22 talk about them.

23 MR. KIM: There again, I'm just for my  
24 sake just trying to get it clear. You said, "What we



1 want to know is whether the finding at Steubenville are  
2 analogous to Illinois." When you say the findings at  
3 Steubenville are analogous, you have to be comparing  
4 that to some findings in Illinois, I assume, so are you  
5 referring to something specific, or are you saying can  
6 the findings in Steubenville be transferred to Illinois?

7 MR. RIESER: I'm assuming this  
8 testimony he has presented had some meaning to the Board  
9 as far as what the study means and the decisions they  
10 should make with respect to Illinois, so I do think we  
11 have to decide how applicable these findings are to  
12 Illinois.

13 MR. KIM: I agree. The only reason  
14 I'm saying that is he's already provided testimony that  
15 he believed the findings could be applied, not only to  
16 Illinois, but to other locales, once you take individual  
17 variances into account. With this question on its own  
18 is read, it seems like you are referring to something  
19 specific, some kind of fact-specific situation or --

20 MR. RIESER: I'm happy to break it  
21 down if the question, itself, is not clear. I'm happy  
22 to break it down and go through each of the issues that  
23 I had in mind and maybe that will take care.

24 MR. KIM: Let's do that.

1 MR. AYRES: I thought we had a  
2 discussion that went for, at least, a half an hour.

3 MR. KIM: I think, if you break it  
4 down, my guess is some of that stuff has already been  
5 answered, but if you want to break it down, that's fine.

6 MR. AYRES: There was a discussion  
7 that was very fact specific, and then we talked about  
8 the transferability of the learning, if you will, from  
9 Steubenville. Do you recall that?

10 MR. RIESER: I recall that and my  
11 problem is my recollection is -- my recollection is that  
12 the answer wasn't clear. There are, obviously, some  
13 informational issues that are portable and Steubenville  
14 was a very specific place that has the conditions that  
15 are represented because of conditions that are specific  
16 to Steubenville.

17 MR. KIM: Sure.

18 MR. RIESER: Seriously, I think we  
19 could have got through this pretty quickly.

20 MR. KIM: I wanted to be clear.

21 MR. RIESER CONTINUES:

22 Q. I honestly don't mean to belabor the  
23 point, but I just don't think it was clear on the  
24 record. Let me ask, Steubenville is in a river valley.

1 Is that correct?

2 A. The city of Steubenville has part of its  
3 boundaries that fall in the Ohio River Valley, that's  
4 correct.

5 Q. Was the sampling location in the part of  
6 Steubenville that was in the River Valley?

7 A. You get down to, I think, definitions. I  
8 think, technically, that whole area is in the Ohio River  
9 Valley, but just to be clear, in terms of topography,  
10 the site was not down on the river in the valley. It  
11 was actually up on top of the area that then flattens  
12 out going to the west in Ohio, so it was not down in the  
13 valley where local sources would have inundated the data  
14 that we were getting. It was actually up on top, and I  
15 think, like, 400 feet above the river, to be clear.

16 Q. Is the meteorology of the Ohio River  
17 Valley, as you have described it, different than the  
18 meteorology one would typically see in Springfield,  
19 Illinois?

20 MR. KIM: I think this has been  
21 answered. I think he went through an extended  
22 discussion about meteorological distinctions between  
23 Steubenville and other parts of any other part of the  
24 country. I think his specific testimony was you are not

1 going to find that replicated anywhere, but  
2 Steubenville.

3 MR. RIESER CONTINUES:

4 Q. Is that your testimony?

5 A. I think what I said yesterday was that the  
6 conditions at Steubenville, in terms of the overall  
7 source-receptor relationships, which meteorology is a  
8 part of, are specific, to a point, and that those would  
9 be unique to Steubenville. The question is whether the  
10 conditions in Steubenville make it anomalous, or somehow  
11 unique where the results would not be transferable to  
12 somewhere else in Ohio, somewhere else in the Great  
13 Lakes or in Illinois and I would suggest that the  
14 conditions are not unique or anomalous to make them so  
15 they are not usable or transferable to conditions that  
16 we would have in Illinois.

17 Q. What are the conditions that are not  
18 anomalous?

19 A. The weather in Ohio, just as it is in  
20 Indiana, Illinois, Michigan, Minnesota, Wisconsin, are  
21 all controlled by synoptic meteorology, which is the  
22 large scale movement of the highest and low pressure  
23 systems across the Great Lakes, so unlike some places  
24 where they have some dominant feature, such as Bermuda

1 high, which may dominate the weather in a certain  
2 location, that might make it different than another one.  
3 The Great Lakes, themselves, are dominated by synoptic  
4 meteorological transport, and therefore, as long as  
5 you're taking into account the specific meteorology that  
6 occurs at that spot, it's not going to be all that much  
7 different. The controlling factors are not that much  
8 different for Illinois than Ohio.

9 Q. Do the presence of a large number of power  
10 plants surrounding Steubenville, does that factor make  
11 it anomalous to other locations?

12 A. Again, I think I said this before, but I  
13 believe that the high density of power plants in the  
14 Steubenville area is a prime reason why power plants are  
15 a prime contributor to the extremely high deposition  
16 that we see there. However, in another place which  
17 would have a similar density, I believe you would see a  
18 similar result.

19 Q. So to see a similar result, you would need  
20 a similar density of power plants?

21 A. You could get similar amounts of  
22 deposition different ways, but if you want to ensure  
23 that you saw elevated, like, 70 percent contribution of  
24 the power plants, you would need to have a significant

1 emissions, upwind emissions, in the local vicinity.  
2 Again, the upwind vicinity, to be clear, of the receptor  
3 model or the receptor site.

4 Q. When you use "upwind vicinity" in that  
5 sentence, what, specifically, do you mean?

6 A. The greater region.

7 Q. What is "the greater region"?

8 A. The local and regional area surrounding  
9 the receptor.

10 Q. The local and regional areas we defined --

11 A. Previously.

12 MR. ZABEL CONTINUES:

13 Q. Dr. Keeler, what is the closest coal-fired  
14 power plant to your monitoring site?

15 A. I believe it's the Samis plant located  
16 about seven kilometers north of the Steubenville  
17 location.

18 Q. Is the Samis plant on the river?

19 A. Yes, sir.

20 Q. What's the height of the stacks of the  
21 Samis plant? Do you know?

22 A. It's fairly tall. I could look it up, if  
23 you would like me to, but my guess is it's 400 or 500  
24 feet.

1 Q. But it's on the river?

2 A. Yes.

3 Q. So it's 400 feet lower than your  
4 monitoring site?

5 A. The base of the plant is 400 feet lower,  
6 and because it's seven miles up the river, I actually  
7 don't know if it's the top of the topographic change,  
8 which would make it more than 400 feet.

9 Q. Seven miles or seven kilometers? There is  
10 a slight difference.

11 A. I believe it's seven miles.

12 Q. Was there any conversation of plume impact  
13 from the Samis plant?

14 A. We actually looked for plume impact on the  
15 Samis plant in our ambient data. We have not modelled  
16 specific events, but we believe that we see, again,  
17 plume impacts at our site. We haven't definitively  
18 looked and said, "This is the Samis plant," but we do  
19 see indications of a coal-fired utility to the north of  
20 our site impacting the levels of mercury in SO<sub>2</sub> and  
21 other things that we have seen there.

22 Q. North of the plant would not be on the  
23 river, I take it? I'm not familiar with the geography.  
24 I'm just asking.

1 A. Yes, it is.

2 Q. So it would be lower than the monitoring  
3 site, as well?

4 A. The height of the stack might be  
5 comparable in height to the elevation of the monitoring  
6 station.

7 Q. How far away is that plant?

8 A. I'm sorry?

9 Q. You said same as the north?

10 A. Yes.

11 Q. I thought you identified the second plant.

12 A. No.

13 Q. We were just talking about Samis?

14 A. That's correct.

15 MADAM HEARING OFFICER: Question H,  
16 then.

17 DR. KEELER: "Is this high density of  
18 coal-fired units reflected in the CMAQ modeling  
19 performed by U.S. EPA?" Again, basing my answer  
20 completely on the figure that you can see in the Exhibit  
21 30, the one we were just referring to, the base case, I  
22 would say yes. Question I: "Is it reflected in the  
23 TEAM modeling performed by AER?" My answer is I don't  
24 have a recollection of seeing that output. I may have



1           seen it. I just don't recall whether it was reflected  
2           or not.

3           Q.       Was it part of the literature that you  
4           reviewed and you talked about reviewing model an --

5           A.       I have read several papers in the  
6           peer-reviewed literature that described and talked about  
7           TEAM model and so forth. I just don't recall seeing or  
8           visually cueing in on some area that had a specific  
9           impact from a specific source type.

10                   MADAM HEARING OFFICER: Mr. Rieser, I  
11           believe that you asked earlier I got the answers to J  
12           and K on the comparison of the Steubenville research  
13           with the CMAQ.

14                   MR. RIESER CONTINUES:

15           Q.       That's correct, but I have one follow-up  
16           on that, which is sort of the opposite. Do you know  
17           what steps the U.S. EPA has taken to compare the results  
18           of CMAQ to your work in Steubenville and what their  
19           findings were?

20           A.       Only, again, what you can see presented in  
21           Landis presentation, that is the only thing I'm aware  
22           of.

23           Q.       Are you aware of the response to  
24           significant public comments received in response to the

1 revision of the December, 2000, regulatory findings on  
2 the emissions of hazardous air pollutants from the  
3 utility, electric utility, steam generating units, and  
4 the removal of coal- and oil-fired -- the  
5 reconsideration technical support document that came  
6 out?

7 A. I got an E-mail as an announcement from  
8 someone saying that this report came out. I have not  
9 had a chance to, either, down load it, or read any of  
10 the pages.

11 Q. So you haven't read the response that says  
12 the results of the Steubenville, Ohio, receptor modeling  
13 study corrected by EPA, ORD, Office of Research and  
14 Development, that Dr. Landis -- are consistent with  
15 those entangled by the CMAQ modeling?

16 A. Okay.

17 Q. You have read that?

18 A. No, I have not.

19 Q. So you haven't had a chance to review why  
20 they say that and determine a response?

21 A. No. I have not downloaded that report or  
22 read any portion of it.

23 Q. Thank you.

24 MADAM HEARING OFFICER: L.

1 DR. KEELER: May I make a follow-up  
2 comment? I think it's important to know that my modeling  
3 half compares to my measurement half. The idea of a  
4 good comparison is a different definition, and so in the  
5 EPA case, without having read what they actually say,  
6 they might view a 43 percent contribution based on the  
7 2001 year to be fairly similar to a 70, plus or minus, a  
8 14 or 15 percent contribution from the receptor  
9 modeling, so just to know how they interpret "good" and  
10 the words that you used for that. I know that they are  
11 trying to put the best fit on that as they can.

12 MADAM HEARING OFFICER: Mr. Rieser,  
13 what exactly were you reading from?

14 MR. RIESER: I was reading from the  
15 Technical Support Document, which is attached to the  
16 reconsideration of the CAMR, mercury CAMR that the EPA  
17 announced on June 9 of 2006. It was in the Federal  
18 Register on that date, which would be 71 Fed Reg. 33388  
19 through 333402. I'm not sure if that's -- I have just  
20 been advised that this is the Federal Register for the  
21 actual reconsideration discussion, which actually  
22 contains a little bit of this in there. The response to  
23 comments which I read is not in the Federal Register.  
24 It's on the U.S. EPA website, so I will be happy to

1 provide a copy to you.

2 MADAM HEARING OFFICER: Thank you very  
3 much. For the record, obviously, since this isn't the  
4 federal register it's a public document and obviously,  
5 we, at the Board, look at the Federal Register, but for  
6 purposes of the record, it's probably best if we also  
7 put the Federal Register reconsideration of CAMR in the  
8 record at the same time. Do you have a clean copy of  
9 that?

10 MR. RIESER: I have clean copies of  
11 either one, so if I may, if I could bring some copies  
12 Monday --

13 MADAM HEARING OFFICER: That would be  
14 great. Thank you.

15 DR. KEELER: Are we -- K is gone.  
16 "What steps have you taken to compare the results of  
17 your Steubenville work with the EPRI TEAM deposition  
18 models, which was included in the CAMR docket?" Again,  
19 this was not the scope of work, nor the scientific  
20 objective of the project that I performed, so it was not  
21 done. Obviously, M is not applicable. That finishes  
22 Ameren's questions.

23 MR. KIM: I asked for a short break  
24 because when we might have neglected to send him the

1 four questions that were presented by Prairie State and  
2 I think he confirmed that he hadn't seen them, and I  
3 think some of them may have been answered, but I want to  
4 make sure that we get all the questions answered.

5 MADAM HEARING OFFICER: I think  
6 Question No. 1, "Have the details of Steubenville model  
7 been made publicly available?" That's similar to the  
8 Ameren -- quite a bit of discussion. Second, "You state  
9 that 70 percent of the mercury wet deposition in  
10 Steubenville comes from coal-fired power plants. How  
11 far have you traced back power plant plumes to reach  
12 that conclusion?" Hundreds of miles, and we also  
13 discussed that today.

14 DR. KEELER: Much longer than that.

15 MADAM HEARING OFFICER: So I think  
16 question No. 3 you might want to elaborate.

17 DR. KEELER: "Have you modeled what  
18 wet mercury deposition is predicted in Steubenville  
19 after implementation of CAMR? I actually just answered  
20 this question to some extent. Our model does not have  
21 predictive capability, so we are not able to do that.  
22 So the rest of the question is not applicable and 4:  
23 "Would you expect coal-fired power plants to contribute  
24 70 percent of the mercury to wet deposition at every

1 location this the United States?" One of the things I  
2 think would be worthwhile just pointing out -- the  
3 answer is no, and one of the things that's worth  
4 pointing out is when you hear these estimates of 8 to 10  
5 percent of the mercury deposition in United States is  
6 from U.S. sources it's kind of a small number. They are  
7 talking about the entire land area of the United States.  
8 Every one single square area and most of the coal-fired  
9 utilities are in the Eastern United States, the eastern  
10 one-third of the United States, and you would not expect  
11 to see 70 percent contribution to Southern New Mexico,  
12 so obviously, you would not expect that. Continuing  
13 with that question, if not, what is the contribution of  
14 the Illinois coal-fired power plants to mercury wet  
15 deposition in Illinois?" I'm not sure exactly how the  
16 questions are connected, but again, I don't have a  
17 quantitative estimate for that to answer that question.

18 MADAM HEARING OFFICER: Anything  
19 further?

20 MR. RIESER: Just to note for the  
21 record that Mr. Bonebrake advised me that he had, in a  
22 fit of preparation, actually brought copies of the  
23 reconsideration for the federal register, and so we will  
24 present those now. Giving the first copy to Mr. Harley.

1                                   MADAM HEARING OFFICER: I will mark  
2 this. This is the actual Federal Register from June 9,  
3 2006. It's the reconsideration. It's not the comments.  
4 We will mark this as Exhibit 31, if there's no  
5 objection. Seeing none, it's Exhibit 31.

6                                   (Exhibit No. 31 was admitted.)

7                                   MR. FORCADE CONTINUES:

8                                   Q. Yes. If I could, I would like to ask a  
9 Steubenville question. If I could, I would like to  
10 direct your attention to three documents in the record  
11 and sort of lay a frame work for the questions. The  
12 first one is Ms. Willhite's testimony on page 3. Why  
13 don't you grab that. In particular, in that document on  
14 the middle paragraph contribution from point sources,  
15 there was a sentence, "It was determined that the total  
16 of all waste water discharges to receiving rivers and  
17 streams in Illinois provide an average annual loading of  
18 45 pounds of mercury." Do you see that?

19                                  A. I do.

20                                  Q. The second would be --

21                                   MADAM HEARING OFFICER: For the  
22 record, Ms. Willhite's testimony is Exhibit No. 8.

23                                   MR. FORCADE CONTINUES:

24                                  Q. Would be the Technical Support Document at

1 pages 68 and 69. On the bottom of 68 and top of 69 in  
2 the reference to total mercury loading from MPTS  
3 sources, and the third one would be your prepared  
4 testimony at page five, the last paragraph, in which you  
5 talk about the importance of coal-fired power plants to  
6 the loading of mercury in large lakes and many down  
7 inland lake which is have been identified as impaired  
8 waters. The distinction I'm trying to suggest here is we  
9 have talked a great deal about deposition, but we have  
10 not particularly identified as much background  
11 information on loading to the lakes and streams, and I  
12 would like to ask a few questions, if I could, to  
13 explore that.

14 First, would you say that past and  
15 present loading to lakes and streams, particularly  
16 impaired streams, is important in determining the amount  
17 of mercury available for methylation?

18 A. Yes. It is important, and I would like to  
19 add that some of the more recent research that had been  
20 performed through the halogen project (phonetic) and  
21 other work which has been published by Cindy Gilmore  
22 suggests that it's the mercury that's recently deposited  
23 from the atmosphere that is the most important in terms  
24 of cycling and methylation and contamination of the



1 ecosystem.

2 Q. I'm still trying to restrict right now my  
3 questions to the loading to streams. You, again,  
4 mentioned deposition.

5 A. Because deposition is the primary input to  
6 most streams.

7 Q. You're running a bit of ahead, if you  
8 wouldn't mind. Would you identify what you would  
9 consider to be the sources of loading of mercury to  
10 impaired lakes and streams?

11 A. Well, there's tributary inputs that load.  
12 There's nonpoint sources from agricultural and animal  
13 feed lots. All of the runoff from industry sites,  
14 basically, runoff from the land into tributaries and  
15 then direct discharges from point sources into either  
16 tributaries or inputs to those lakes and streams.

17 Q. Have you done any studies to try and  
18 evaluate those specific forms of loadings and quantify  
19 them?

20 A. Anywhere?

21 Q. Yeah. Let's start with anywhere.

22 A. Yes.

23 Q. Could you tell me what studies you have  
24 done?

1           A.     We did a rather large study in the city of  
2     Detroit trying to -- not trying, with the objective to  
3     quantify the importance of atmospheric deposition and  
4     the goal of trying to identify the most significant  
5     sources of mercury and PCB's to the receiving waters  
6     that feed into the city of Detroit's waste water  
7     treatment facility.

8           Q.     I'm sorry. Was that loading to the sewers  
9     and waste water treatment facility?

10          A.     Yes. In other words, we worked with the  
11     city of Detroit's waste water -- the water and sewer  
12     department to quantify how much mercury was coming into  
13     the head of their waste water treatment plant from all  
14     of their network, which is a fairly large network of  
15     homes, industry, combined storm sewer overflows and so  
16     forth, and then to look at how much mercury they then  
17     discharged, and part of this was wrapped around their  
18     MPDS permit looking at how much affluent mercury they  
19     put out and what forms, and how far of this was related  
20     to atmosphere I think deposition.

21          Q.     So you have done studies on the amount of  
22     and source of mercury entering into a public-owned  
23     treatment work?

24          A.     Yes.

1 Q. Have you done similar studies for a stream  
2 or lake?

3 A. We have not done an exhaustive study in  
4 trying to estimate the inputs to a specific tributary.  
5 We have done some monitoring in the state of Michigan  
6 trying to look at kind of upstream-downstream  
7 relationships for mercury on rivers in Michigan, but  
8 again, those were not meant to be exhaustive, in terms  
9 of, specifically, quantifying the runoff from  
10 agriculture, versus industrial discharge. It was just  
11 trying to see if we could find an influence of an urban  
12 area on a specific river or tributary.

13 Q. Could you identify the number or name of  
14 those studies for me, so I can ask some questions on  
15 them?

16 A. The one was -- I don't know what the  
17 initial name was, but it was something like "Mercury  
18 Levels in Michigan Rivers and Surface Waters" done with  
19 the Michigan DEQ in the 90's. I actually don't remember  
20 the exact year, sometime in the 90's, though, and then  
21 the other study we did was something mercury PCB's and  
22 cadmium (phonetic) in affluent -- at the city of  
23 Detroit's waste water treatment plant, something like  
24 that.

1 Q. The second one you mentioned, would that  
2 be the POTW study you discussed earlier or was that a  
3 third?

4 A. No. That's the same one.

5 Q. So excluding the publicly-owned treatment  
6 works, as far as streams lakes and rivers are concerned,  
7 you have one study which was the 1990 Michigan study.  
8 Is that correct?

9 A. Where we look, specifically, at streams  
10 and rivers.

11 Q. The loading of mercury to streams and  
12 rivers.

13 A. Again, as I mentioned, it wasn't a study  
14 looking at -- it was looking at levels of mercury kind  
15 of upstream and downstream of different urban areas or  
16 towns, so that would be my only experience.

17 Q. Regarding the 1990 -- do you want to call  
18 it study or do you have another term?

19 A. Which one are you referring to?

20 Q. The 1990 Michigan --

21 A. Michigan study. That's fine. When you  
22 are asking me that, you are not asking me to talk about  
23 the Lake Michigan Mass Balance Study.

24 Q. No. I'm asking you what you have done to

1 evaluate the mercury loading to streams and rivers.

2 A. That's not my main focus area, in terms of  
3 research.

4 Q. I understand, so the 1990 Michigan study  
5 was the only one that you can point to, specifically.  
6 Is that correct?

7 A. I guess, yes.

8 Q. Could you describe in detail what you did  
9 in that study?

10 A. We made measurements of -- well, I should  
11 say the Michigan DEQ took water samples using the  
12 systems that we developed to measure total and dissolved  
13 mercury and other metals in a series of rivers, and I  
14 believe maybe lakes and across the lake and in  
15 situations where they could make a measurement, and say,  
16 upstream of Ann Arbor and downstream of Ann Arbor in the  
17 Huron River, for example, to see if there's an influence  
18 of the runoff and inputs of metals and mercury into the  
19 stream related to that area, so it was a way to get some  
20 survey data across the state.

21 Q. When you say the Michigan Department of  
22 Environmental Quality samples, these were water quality  
23 samples?

24 A. Water quality, only.

1 Q. Water quality, only. Do you have an idea  
2 of, approximately, how many sites were studied or  
3 samples were taken?

4 A. I don't recall the total numbers. It was  
5 in the hundreds.

6 Q. Was it a large geographic area or was it a  
7 single stream segment?

8 A. I know that it encompassed the entire  
9 lower peninsula.

10 Q. Of Michigan?

11 A. Yes.

12 Q. That would be a big study, then.

13 A. Yes, sir.

14 Q. What was your function, then, to take the  
15 analytical results from Michigan DEQ and evaluate them  
16 in some way?

17 A. No. Our lab that was doing that provided  
18 the sampling equipment and did all the analytical  
19 determinations of mercury and metal concentrations in  
20 the samples collected and provided them with the  
21 recorded findings. Because that's outside my typical  
22 area of research, I didn't spend any time evaluating the  
23 data, other than for the quality of the information that  
24 was collected.

1           Q.     Did you draw any conclusions from the data  
2           or was your evaluation simply, "Here's the analytical  
3           results"?

4           A.     As you had described, "Here's the  
5           analytical results."

6           Q.     So you performed no evaluation of the  
7           sources or impacts or --

8           A.     No, sir.

9           Q.     Have you done any evaluations of the  
10          loading of mercury to stream segments, other than this  
11          that would be relevant to determining the amount of  
12          mercury coming into the lakes and streams?

13          A.     I think I have said that my expertise is  
14          not in tributary or aquatic science.  It's in  
15          atmospheric science, so I have not performed input  
16          analysis.

17          Q.     Do you have any method or are you aware of  
18          any additional reports that have been done which  
19          evaluate the amount of loading to specific stream  
20          segments of mercury comparing, for example, air  
21          deposition, versus upstream water deposition?

22          A.     Well, there was an extensive amount of  
23          work done by the University of Wisconsin and the USGS,  
24          as part of the Lake Michigan Mass Balance study in

1 looking to -- I don't know how many tributaries into  
2 Lake Michigan, and I know they have done an exhaustive  
3 amount of work for Lake Superior for exactly that same  
4 purpose for mercury and trace elements, and I have seen  
5 presentations, and I know there's, at least, a few  
6 publications in the peer-reviewed literature on that.

7 Q. The analytical results that you conducted  
8 for the Michigan study, those are water quality  
9 evaluations for mercury water analysis for mercury  
10 content?

11 A. Yeah. They were analysis of liquid  
12 samples that were taken from surface bodies of water.

13 Q. When were those done?

14 A. I said I believe it's in the mid 90's. I  
15 don't recall the exact date.

16 Q. Do you happen to recall what analytical  
17 method you used to test the mercury?

18 A. Sure. I used coal vapor atomic  
19 fluorescence for the mercury and I used ion-coupled  
20 masstometry (phonetic) for the trace element analysis.

21 Q. So would that be Method 1631, Provision E  
22 of the mercury analysis?

23 A. That protocol came out after, but our  
24 protocol is almost identical. I mean, we did not follow



1           their protocol. We have our own protocol that has been  
2           subjected to Agency peer review, and proved quality  
3           assurance, quality control plans that we use in all of  
4           our work.

5           Q.     What was your method detection limit?

6           A.     For that study, I would have to go back.  
7           I don't recall the detection limits for the 40-some  
8           elements that we did.

9           Q.     I'm sorry, restricting my evaluation here  
10          to mercury.

11          A.     Again, I don't recall, but it was for  
12          that -- we determine a method detection limit for every  
13          study we do, based on the actual data, which is defined  
14          as seven times the standard deviation of repeated  
15          analysis of a low standard, and I don't remember the  
16          exact number. It's -- I believe it's around a tenth of  
17          a nanogram per liter, could be lower than that.

18          Q.     Prior to the 1990 Michigan study, did you  
19          do other evaluations of mercury concentrations in water?

20          A.     I did determinations of mercury in cloud  
21          water and fog water, yes.

22          Q.     What time period would that be in?

23          A.     Late 80's.

24          Q.     Did you use the functional equivalent of

1 1631 for those tests, too?

2 A. No. We used a much more elaborate and  
3 exhaustive technique that requires a nuclear reactor. I  
4 did this work at the Institute of Technology while I was  
5 visiting scientists there.

6 Q. Prior to 1990, if you did not have access  
7 to a nuclear reactor, would you not have commonly used  
8 Method 245 or a similar method for determining mercury  
9 content in waters?

10 A. I'm not sure I can answer that question.  
11 I'm not 100 percent certain I know what "Method 245" is,  
12 and I know that -- well, I'm not sure I can answer that  
13 question.

14 Q. If you had conducted a water analysis for  
15 mercury prior to the 1990's, and prior to the  
16 implementation of your test protocol, do you know,  
17 approximately, what the method detection limit would  
18 have been available for those prior test methods?

19 A. If you are talking about research methods  
20 or are you talking about those that would have been used  
21 by states or EPA? I mean --

22 Q. States and EPA and facilities subject to  
23 mercury testing.

24 A. You are asking me to give you a historical

1 account of the analytical techniques through the 80's?

2 Q. That would be one way of getting to my  
3 question, but another way would be to say is it your  
4 understanding that Method 1631 is described as being,  
5 approximately, 250 times more sensitive than the prior  
6 testing protocols for mercury with significant increase  
7 in lowering of the method detection limit?

8 A. I don't know the numerical number, but  
9 from the early 1980's, the methods that were used such  
10 as atomic absorption and coal vapor atomic absorption  
11 have detection limits that were much higher, so they  
12 were unable to see the small quantities of mercury that  
13 we can see, starting in the late 80's. For some reason,  
14 they were already beginning to use those low-level  
15 techniques in the early 80's. Nick Bloom and Bill  
16 Fitzgerald and his students at the University of  
17 Connecticut had these techniques at their disposal. The  
18 widespread use of those were regulatory, and other  
19 monitoring purposes did not happen in the 80's.

20 Q. I believe it would be 1990 would be I  
21 think the first -- do you happen to recall when the  
22 first version of Method 1631 came out or its equivalent,  
23 which uses the nanogram detection limit?

24 A. I don't know the exact date because I

1 restarted using something that was similar before that  
2 method and it actually worked its way through the  
3 system.

4 Q. Would you believe it would be common,  
5 prior to that time, for the detection limit to have been  
6 about one part per billion or slightly less for most of  
7 the common analytical procedure in use by states and  
8 dischargers?

9 A. Well, my understanding, again -- and I  
10 can't say this is for most or I can't put a quantitative  
11 number on it, but I would say that most places that were  
12 doing fish contaminant work and so forth had, at least,  
13 a detection limit of 150 parts per trillion maximum  
14 before they moved to coal vapor atomic fluorescence and  
15 some of the other techniques, so it was quite a bit  
16 higher than the .1. I mean, that's a thousand fold  
17 different than I just quoted you 150 parts per trillion  
18 versus the .1 that we are able to do now.

19 Q. But that was for --

20 A. Total mercury.

21 Q. Tissue evaluation?

22 A. Mercury analysis. Are you asking me for  
23 surface water?

24 Q. All my questions relate to mercury testing

1 for surface water.

2 A. I don't know the exact number to give you,  
3 but your basic premise is correct, that detention limits  
4 were very high and could have been approaching a part  
5 per billion prior to the advent of the work that  
6 Fitzgerald and his students did in the early 80's.

7 Q. If you were to evaluate the loading to a  
8 stream, as you mention on page five of your testimony,  
9 how would you determine the components, other than air  
10 deposition?

11 A. Again, this is outside my area of  
12 expertise, but I did participate in the Lake Michigan  
13 Mass Balance study and from working collaboratively with  
14 the University of Wisconsin and the USGS and people who  
15 did that work, as I understand it, they make  
16 measurements of the hydrograph, the flow of water coming  
17 in from all the major tributaries they think are  
18 important, and then try to take samples for mercury  
19 using approved clean techniques to then ascertain what  
20 the amount of mercury coming in along at different flows  
21 from that tributary, and then sum those numbers up, so  
22 they are a flow-proportioned calculation of how much  
23 mercury would enter into the body of water that you are  
24 interested in and studying. That's, more or less, what

1       they did for all the tributaries for Lake Michigan. In  
2       doing that, you come up with a total mass that entered  
3       Lake Michigan from the Sheboygan River, and you do that  
4       for every single one of them, so it requires an enormous  
5       amount of work and a lot of monitoring.

6               Q.     If you were to evaluate, for example, an  
7       impaired lake or stream in Illinois in an attempt to  
8       determine the amount of air deposition in loading to  
9       that stream and compare it to the loading from other  
10      sources, what would you do for the air deposition  
11      portion to determine loading of that stream?

12             A.     You are asking me, personally, what I  
13      would do?

14             Q.     Yes.

15             A.     If it was a specific stream that I was  
16      interested in, I first have to evaluate how long that  
17      stream was, whether I felt that one monitoring location  
18      would represent what would be coming into that stream if  
19      it was extremely long, 20 miles long, or if it's  
20      something longer than that, I would evaluate what point  
21      sources were contributing to that stream, and then look  
22      at the land use types around that stream, and then make  
23      a determination whether I needed to put one or two  
24      deposition where I would collect the amount of

1 precipitation that fell and I gauge it, so I would have  
2 a number of rain gauges and so forth along there, so I  
3 could get an accurate description of the amount of  
4 precipitation that fell into the stream and maybe even  
5 put multiple gauges on that stream, so I could get the  
6 flow of that stream, so I could have a good idea of then  
7 the mass of whatever contaminant I was looking into the  
8 stream, but it would be very stream specific, and it  
9 would require some analysis of the situation and on-site  
10 recognizance and so forth.

11 Q. If you were to complete such a study and  
12 determine what you felt was an appropriate level of  
13 loading from air deposition, that would simply be one  
14 component of the loading to that particular stream  
15 segment, wouldn't it?

16 A. Yes, sir.

17 Q. And there would be another section at  
18 which would be the water component, sediment component.  
19 Those other components would all contribute to the total  
20 loading?

21 A. I didn't say explicit, but clearly, you  
22 would have to make repeated measurements of the stream,  
23 the water body of interest, both, the liquid sample, as  
24 well as the sediments, anything else you think could

1 have contributed to the loading or movement of the  
2 contaminant through that stream.

3 Q. There would be no way that you could  
4 possibly determine the relative proportion of the air  
5 loading to that stream segment without knowing the other  
6 components of contribution, would there?

7 A. If I was an engineering firm who was  
8 requested to make a determination of the importance of  
9 runoff, atmospheric deposition and so forth, I would use  
10 my best expert judgment and use what data was available  
11 to estimate the importance of these. In many cases you  
12 don't have the actual measurements, so you have to use  
13 expert judgment, and use whatever available measurements  
14 are there to give an estimate for that. In the best of  
15 all worlds, I would like to have measurements. I'm a  
16 very measurement-greedy person. I like to have  
17 measurements for all those things that I said, so I can  
18 be certain of my conclusion. However, in making a best  
19 estimate judgment, one would use the available data that  
20 you had and put some air bars on how precise or  
21 imprecise you would be able to estimate the various  
22 inputs.

23 Q. Directing your attention to page three of  
24 Ms. Willhite's testimony and page 69 of the Technical



1 Support Document, in which it suggests that the average  
2 MPDS has loading to Illinois streams was 45 pounds and  
3 then comparing that to 7,022 pounds per year of mercury  
4 emissions, that evaluation wouldn't give you any way of  
5 determining for an impaired stream what portion was  
6 coming from air deposition and what portion was coming  
7 from other sources, would it?

8 A. Those two facts by themselves do not allow  
9 you to calculate the specific proportion coming from the  
10 atmosphere, versus what was coming from the tributary to  
11 a specific stream.

12 Q. Would a portion of the reason be because  
13 it talks about emissions to the atmosphere, rather than  
14 amount loaded to the Illinois streams from air  
15 deposition?

16 A. I think that's one source of uncertainty,  
17 yes.

18 Q. There would be no evaluation in the  
19 sentence that you see there about other contributing  
20 sources, such as sediment movement, nonpoint runoff,  
21 combined sewer overflows, types like that?

22 A. I'm sorry. I don't know where you are  
23 referring to.

24 Q. The reference to the 40 pounds of MPDS

1 loading compared to the 7,022 pounds of air emissions,  
2 it makes no reference to any contribution coming from  
3 surface runoff from sedimentation moving downstream or  
4 other sources, does it?

5 A. I don't see any listed here.

6 Q. And would you consider that to be one  
7 possible input to the stream loading for an impaired  
8 water?

9 A. Yes. There is one potential input to a  
10 potentially impaired water, yeah.

11 Q. Would you have any information that would  
12 allow you to draw a rough conclusion of the amount of  
13 stream loading to a particular stream segment that would  
14 come from air deposition, versus other sources,  
15 generally?

16 A. I wasn't asked to look at this or address  
17 that, and that's not, again, in my typical line of  
18 research. If I was asked to do something like that, I  
19 may be able to put that information together, but I did  
20 not do that, and this is the first time I have actually  
21 thought able calculating something like that. I mean,  
22 we have looked at this issue, in terms of looking at one  
23 of the things we found on the Lake Michigan Mass Balance  
24 Study was there a fairly large amount of the input to

1 tributaries that they couldn't account for by looking at  
2 runoff from agricultural areas or runoff from surfaces  
3 and so forth, which we did a rough calculation and  
4 determined that most of those was probably from  
5 atmospheric, so it was the mercury that was in the rain  
6 that fell to the ground, and then wound up running off  
7 into the tributary. It was in the soluble phase, and so  
8 forth, but that kind of gets at what you're asking me  
9 here, but I have not addressed that question,  
10 specifically, in this case.

11 Q. Right, but what you're talking about there  
12 is mercury entering in the equatous (phonetic) phase.  
13 Is that correct? You're not talking about --

14 A. Yeah. The river has water in it and the  
15 rain fell and fed and goes as a liquid into the river,  
16 yeah.

17 Q. So for the evaluations that you were doing  
18 in Michigan, and for the discussion you just had, would  
19 you use filtered samples to determine the mercury  
20 content?

21 A. If I wanted to understand the mechanism  
22 and the physiochemical transport from various fields and  
23 so forth, I would take a total sample and definitely  
24 filter it, so I could look at the total and dissolved

1 phase, yes.

2 Q. And historically --

3 MADAM HEARING OFFICER: I apologize  
4 for interrupting, but are you going to be able to wrap  
5 this up? I don't want to cut you short, but it's 10  
6 after 12, and we've been back at for a couple hours, so  
7 if it's going to take another 20 minutes or so we might  
8 want to go ahead and take a break, but if it's only  
9 going to take a couple -- it's up to you.

10 MR. FORCADE: I'm not sure whether  
11 it's going to take me five or 15. I'm not sure.

12 MADAM HEARING OFFICER: Let's go ahead  
13 and go to lunch because we have been back at it for a  
14 couple of hours.

15 MR. FORCADE: That's fine. I just  
16 don't know.

17 (At which time, the proceedings were  
18 adjourned for a lunch break.)

19 MADAM HEARING OFFICER: I believe  
20 Mr. Forcade was asking Dr. Keeler some questions. Let's  
21 go back.

22 MR. FORCADE CONTINUES:

23 Q. We had a lengthy discussion at lunch about  
24 the co-benefits discussion about the co-benefits of

1 concluding the testimony as early as possible on a  
2 Friday afternoon, so I may be able to conclude with one  
3 or two more questions. Dr. Keeler, if I'm not  
4 characterizing this correctly, please let me know, but I  
5 believe that you've made attempts to identify the source  
6 of mercury deposition by what is, essentially, a  
7 fingerprinting to identify the source categories. Is  
8 that a paraphrase?

9 A. No. That's correct.

10 Q. Have you done anything to try and identify  
11 the mercury present in fish in a similar manner?

12 A. I, personally, have not.

13 Q. That's it.

14 MR. HARRINGTON CONTINUES:

15 Q. One question. The City of Detroit Study  
16 that you did concerning BOTW's, do you know whether  
17 there was any effort to characterize, either bypasses or  
18 combined sewer overflows during that study and their  
19 impact on receiving waters.

20 A. Our study was focused completely on  
21 quantifying the importance of atmospheric deposition on  
22 PCB's, mercury to the waters that hit the head of the  
23 plant, and we didn't really look at issues of bypass or  
24 combined sewer overflows. We made measurements in the

1 system, but we did not look at those issues.

2 Q. Was there significant measurable mercury  
3 in the influence of the BOTW's?

4 A. We measure mercury in every sample we  
5 collect, and yes, there was measurable mercury. Again,  
6 our detection limits are a tenth of a part per trillion,  
7 so you see mercury in the drinking water you have in  
8 front of you. It's probably not of any concern.

9 Q. To put another way, mercury is everywhere  
10 in the natural environment, correct?

11 A. Yes.

12 Q. Do you recall what the levels of mercury  
13 were in the influence of publicly-owned treatment works?

14 A. The influence concentrations varied from  
15 100 to 400 nanograms per liter.

16 Q. Thank you.

17 MR. FORCADE CONTINUES:

18 Q. Dr. Keeler, you just mentioned in response  
19 to Mr. Harrington's question that you found mercury in  
20 many locations. Would you expect, at that level of  
21 detection, that you would find mercury in many MBTS  
22 discharges (phonetic)?

23 A. I believe I would see mercury at least in  
24 the trace quantities in the part per trillion level in

1 all discharges.

2 Q. Thank you.

3 MR. BONEBRAKE CONTINUES:

4 Q. Just one other follow-up, do you recall  
5 the eruption of Mt. St. Helens around 1980?

6 A. I remember it very well, yes.

7 Q. Was that eruption a significant source of  
8 mercury air emissions?

9 A. Volcanoes are thought to be one of the  
10 prime natural sources that put mercury into the earth's  
11 atmosphere.

12 Q. Has there been any estimate of the amount  
13 of mercury emitted into the air that resulted from that  
14 eruption?

15 A. It's possible someone did a calculation.  
16 I'm not aware of that that number.

17 Q. Have you ever seen a comparison of that  
18 number, whatever it may be, to mercury emissions from  
19 electric-generating units?

20 A. Again, I don't recall Mt. St. Helens, in  
21 specific. I know that people have looked at volcanic  
22 emissions from Italian volcanoes and a couple of others  
23 around the world, but I don't recall Mt. St. Helens,  
24 specifically, but it's a pretty large number. If

1       volcanoes were going off continuously all the time or  
2       every year, it would certainly change the global budget  
3       of mercury in the atmosphere. Mt. St. Helens, if you  
4       remember, gave us very beautiful sunsets and put a lot  
5       of ash and sulfuric acid up in the stratosphere, so it  
6       shot -- not only did it put a huge ash over a small  
7       area, but it also shot a lot of stuff up into high  
8       levels in the atmosphere, so it definitely was seen  
9       everywhere.

10               Q.     In fact, the mercury that went up high  
11       levels of atmosphere, would that suggest what mercury  
12       would have been dropping onto the ground or into the  
13       waters for a significant period of time after the  
14       eruption?

15               A.     Whatever came out of the volcano,  
16       including mercury, would have been emitted into high  
17       altitudes and some of that perhaps could have been  
18       removed, since a large of amount went up in the  
19       stratosphere. I'm sure a great deal of it is probably  
20       still up there.

21               Q.     Thank you.

22                               MR. HARLEY CONTINUES:

23               Q.     At the beginning of your responses to the  
24       questions that were posed by Mr. Forcade, you began to



1 describe a study -- actually, there are two authors of  
2 that study who were looking at the relative contribution  
3 of atmospheric deposition to recent contribution and  
4 cycling. Do you recall your reference to that study?

5 A. I'm sorry. I must have post-lunch brain  
6 deadness. Help me a little bit.

7 Q. You were talking about the total loading  
8 and methylation issue, and you began to talk about a  
9 study, which I took to mean that recent contributions of  
10 atmospheric deposition were most important, in terms of  
11 methylation and also most important, in terms of  
12 cycling, and I felt that you had more to say about that,  
13 and I would like to hear what you had to say about that  
14 study.

15 A. You're referring to I believe I referenced  
16 Cindy Gilmore and Dave Griminhoff's (phonetic) work, as  
17 well I think Jim Hurley from the University of Wisconsin  
18 in the water chemistry program. Actually, there's a  
19 fairly large group doing some work in METALLICAS, so  
20 there's a couple different studies there, but what the  
21 issue is there have been for years -- in fact, the  
22 thought was that the methylmercury that was entering  
23 into ecosystems and bioaccumulating was coming from the  
24 bottom sediments over time and some of this born out in

1 Florida through some studies, but over time, people  
2 started to say, well, they couldn't explain the amount  
3 of mercury and methylmercury, specifically, and as they  
4 got better and better at making methylmercury  
5 measurements and the BIODA and in the fish and through  
6 the ecosystem, they started to hypothesize that the  
7 mercury that was actually cycling, so going from the  
8 reactive mercury form methylated by the bacteria, and  
9 then going into the organisms was actually the mercury  
10 that was falling out of the sky today, so the stuff that  
11 falls out today gets chemically transformed and actually  
12 winds up -- that's the mercury that winds up  
13 contaminating the fish so Cindy Gilmore and colleagues  
14 have done some studies where they have actually taken  
15 and used as a tracer an isotope of mercury. It's one of  
16 the masses of mercury, and they can get this mercury  
17 from Oak Ridge National Laboratories, and other places,  
18 and they can put that in the system in different  
19 compartments and look to see where it goes, and in fact,  
20 the mercury that they put in just as precipitation would  
21 have been gone into the lake is what they actually are  
22 now seeing coming up from the ecosystem in a fairly  
23 rapid fashion, so their experiments are duplicating what  
24 their hypothesis -- or confirming their hypothesis that



1           A.     METALLICAS, and I'm trying to think of who  
2     the lead investigator was, but if you look up Gilmore or  
3     Griminhoff or Hurley, I think John Rude up in Canada is  
4     a principal in that.  It's a fairly large team.  I know  
5     Steve Linberg and Jim Hurley were also involved.  It's a  
6     team of about 15 different people from, both, U.S. and  
7     Canada that are doing that work.

8           DR. KEELER:  The key is trying to get  
9     at what's most important in terms of contaminating the  
10    environment, and that's what the focus of that whole  
11    study is and the indication is that it's recent  
12    deposition really is the most important.

13          MR. FORCADE:  When the questions are  
14    over, I have a procedural question for Mr. Kim.

15          MR. BONEBRAKE CONTINUES:

16          Q.     I'm just curious.  You just talked about  
17    the relative significant of more recent deposition.  
18    Does the answer change, Dr. Keeler, in waterbodies in  
19    which the sediment is stirred up for reasons maybe  
20    associated with a hurricane in an area or perhaps  
21    seasonal flooding or drudging.

22          A.     It really depends upon how the mercury is  
23    bound in the sediment or in the soil or whatever you are  
24    referring to there.  If it's tightly bound, it may not

1 be biologically available. Mercury that's in sand,  
2 copper sludge that they get from a copper mine, like up  
3 in the UP of Michigan, for example, is not thought to be  
4 very biologically available, so if it's in that form,  
5 it's stirred up it may not actually lead to a higher  
6 methylation rate. If it's in a bioavailable form, then  
7 yes, it could.

8 Q. So would it be your sense that the  
9 question of the relative significance of recent  
10 deposition is going to be a question whose answer is  
11 somewhat variable depending upon the particular  
12 characteristics of a water body?

13 A. It would be characteristic of a frequency  
14 of disturbance and the type of disturbance and the  
15 characteristics of the water body and the inputs, yes.  
16 All those things would be important to take in mind,  
17 take into account.

18 DR. GIRARD CONTINUES:

19 Q. I hate to bring up Steubenville again, but  
20 Dr. Keeler, I seem to recall that you talked about doing  
21 air sampling every hour during that study looking at  
22 mercury levels in the air. Was I correct in that?

23 A. Yes, you were correct.

24 Q. When you had rain events, and you had this

1 sampling going on every hour, did the amount of mercury  
2 in the air change after the rain event?

3 A. Again, we haven't done an exhaustive study  
4 of every single rain event, but for the ones that we  
5 could match up or that we did match up the ambient that  
6 you do see a very rapid dropout in the reactive mercury  
7 and you see a smaller, but significant, drop in the  
8 particulate mercury levels, as well, with a very -- or a  
9 less of a decrease in the elemental mercury  
10 concentrations, so we see that, not just at  
11 Steubenville, but we see that in our mercury sites in  
12 Michigan, as well, both, in Detroit and at the site in  
13 Dexter, which is about 30 miles to the west of Ann  
14 Arbor, so you do see what's in ambient air increasing  
15 precipitating systems.

16 Q. You hate to quantify it because the data  
17 hasn't been fully processed, but just ballpark. I mean,  
18 are we talking just a 50 percent reduction or 100  
19 percent reduction? What kind of a reductions are you  
20 looking at after a rain event?

21 A. If it's a prolonged rain, if it's a rain  
22 that lasts more than an hour because that's hour time  
23 frame of our measurement, so we have to have something  
24 that goes longer than an hour in order for me to answer

1        your question.  If it's a longer rain that goes more  
2        than an hour, you will see complete depletion of the  
3        reactive mercury from the air.  Particulate mercury is  
4        already very low, so that goes to nothing, as well, and  
5        elemental will stay, approximately, with a background is  
6        1.5 nanograms per cubic meter and doesn't really change  
7        much in terms of through a rain event, so after an  
8        event, say, that's over an hour where it reduces the  
9        mercury in the air down to zero, how quickly do you see  
10       the mercury levels go back up?  It really depends upon  
11       the reason that the rain fell.  If it was a frontal  
12       system where the wind is changed, sometime it doesn't  
13       come back up for quite a while.  If the wind stays coming  
14       from the same direction, as soon as the rain is over,  
15       there will be a period of time where it could be like  
16       the next hour or the hour after that you will see it  
17       start to raise back up again, so there is a one-to-one  
18       correspondence.  I just haven't quantified it, but at  
19       our site in Dexter, we actually -- did quantify it for a  
20       period or time and it was exactly a 50 percent reduction  
21       an hour after the rain came, but then we were making  
22       measurements every other hour, so that we couldn't -- it  
23       wasn't as clean.  For a long rain, you see it go right  
24       to nothing.  For a rain that lasted 30 minutes, the

1 following hour reactive mercury would be half as much,  
2 and then whether it came back up or not depended upon  
3 which way the wind was blowing. The reactive and  
4 particulate mercury concentrations that we see at  
5 Steubenville and in Michigan are very wind directional  
6 and very transport specific. We get transport from the  
7 north in Michigan, for example, we see very little often  
8 with strong winds, especially during the wintertime and  
9 in the summertime, when the wins switch around, you see  
10 very high concentrations, but with very specific wind  
11 directions, and it always responds the same when it  
12 comes to precipitation. We do not have any examples of  
13 high RGM with precipitating events going through for  
14 multiple hours where it stays up and that goes along  
15 with the understanding of how soluble and reactive  
16 mercury is in the atmosphere.

17 MR. FORCADE: If I could have a  
18 procedural question. Earlier this week in the panel  
19 discussion I asked a series of questions about MPDS  
20 discharges mercury content and the contribution of total  
21 loading as described on page 69 of the Technical Support  
22 Document. There are two references supporting that in  
23 the Technical Support Document, and both of those  
24 references are incorrect. They have absolutely nothing



1 to do with calculating mercury, affluent limitations.  
2 As far as I'm concerned right now, there is no support  
3 for those tables. I have asked -- and I don't need the  
4 information today -- but there's a significant open  
5 question which I need information from the Agency to  
6 explore, and I would like to just, not for purposes of  
7 asking questions today, just remind them that that is a  
8 significant open question that I need to explore for the  
9 conclusion of these hearings next Friday.

10 MR. KIM: Indeed, Mr. Forcade did  
11 raise that. We tried to initially provide the documents  
12 that were referenced, and I think I am in agreement with  
13 him that it doesn't really seem to mesh up with what was  
14 in there, so we are in the process of getting that  
15 information, and I'm pretty certain we are going to have  
16 it for you Monday or Tuesday, at the latest, next week.  
17 So if it could wait until then.

18 MR. FORCADE: That's fine with me.

19 MR. KIM: It's not forgotten.

20 MADAM HEARING OFFICER: There were a  
21 couple of things today that we had talked about from  
22 Dr. Keeler. One was I believe you have in front of you.

23 MR. KIM: Actually , I was going to  
24 say there were a couple things I was going to bring up

1 just to sort of close some loops. METALLICUS, the  
2 acronym, I can give you what that stands for, if you  
3 want that. And then the last thing -- this was just  
4 something that was referenced after trying to get a copy  
5 of the Powerpoint presentation that Dr. Keeler made  
6 reference to a while back and I think it's -- I think  
7 the Board and the Agency were the only people that  
8 didn't actually have this. I think I have two copies.

9 MADAM HEARING OFFICER: If there's no  
10 objection, we will mark this as Exhibit No. 32. Seeing  
11 none, "Mercury Deposition in the Great Lakes Region,  
12 James Keeler, University of Michigan Air Quality  
13 Laboratories" is marked as Exhibit 32.

14 (Exhibit No. 32 was admitted.)

15 MR. KIM: I believe this is maybe in  
16 more specific with some nice pictures, but it's,  
17 essentially, everything that -- there is a Far Side  
18 cartoon I think it's pretty much a Powerpoint  
19 presentation of everything that's been testified to thus  
20 far, but we did want to make sure we got that in. It  
21 took a little while because it was all in color and we  
22 had technical difficulties. We have no more color ink  
23 left in the building.

24 MADAM HEARING OFFICER: The other item

1 that we discussed this morning was the emissions data  
2 that was a part and a map. Mr. Matoesian was talking  
3 about that this morning and you said you would get that  
4 for us.

5 MR. KIM: Yes.

6 MADAM HEARING OFFICER: Are we still  
7 in the process of looking for that?

8 MR. KIM: That's information that we  
9 are going to have to print out I think from U.S. EPA's  
10 website, so that might -- we'll get that, but it may not  
11 be until Monday or Tuesday.

12 MADAM HEARING OFFICER: That's fine.  
13 I just wanted to double check on that.

14 MR. KIM: We have four copiers on the  
15 Division of Legal Counsel floor. At any given time, one  
16 of them is working, so I'm assuming that ratio holds  
17 true throughout the Agency.

18 MADAM HEARING OFFICER: Is there  
19 anything else for Dr. Keeler?

20 MR. RIESER: Maybe one thing to do.  
21 Obviously, we visually observed the LADCO report at the  
22 time it was given, but we haven't had a chance to look  
23 at it. Perhaps if Dr. Keeler can sit for another hour  
24 or, so maybe we can proceed with Dr. Hornshaw give us a

1 chance to look at this, and then if we can come back.  
2 It shouldn't take that long, but I just need to check to  
3 see if there's any other questions.

4 MADAM HEARING OFFICER: Is that  
5 possible, Dr. Keeler?

6 DR. KEELER: That's about right. I  
7 have another hour and 10 minutes.

8 MR. RIESER: If it would help, the  
9 minute I'm ready to ask some questions, if I have any  
10 one way or the other, I can put my hand up. I hate to  
11 interrupt the flow of somebody else's questioning.

12 MR. KIM: Two airlines in Springfield,  
13 so I think our people are just struggling to hold on to  
14 the reservations they have got, but that's fine. If,  
15 during his testimony you want to just interrupt, that's  
16 fine with us.

17 MADAM HEARING OFFICER: Then shall we  
18 begin with Dr. Hornshaw? Before that, Dr. Keeler I want  
19 to, personally, thank you very much. It's been very  
20 enlightening.

21  
22  
23  
24

1 STATE OF ILLINOIS)  
2 COUNTY OF ST. CLAIR)SS  
3

4 I, Holly A. Schmid, a Notary Public in  
5 and for the County of Williamson, DO HEREBY CERTIFY that  
6 pursuant to agreement between counsel there appeared  
7 before me on June 16, 2006, at the office of the  
8 Illinois Pollution Control Board, Springfield, Illinois,  
9 Dr. Gerald Keeler, who was first duly sworn by me to  
10 testify the whole truth of his knowledge touching upon  
11 the matter in controversy aforesaid so far as he should  
12 be examined and his examination was taken by me in  
13 shorthand and afterwards transcribed upon the typewriter  
14 (but not signed by the deponent, his signature having  
15 been waived by agreement of counsel) and said deposition  
16 is herewith returned.

17 IN WITNESS WHEREOF I have hereunto set  
18 my hand and affixed my Notarial Seal this 30th day of  
19 June, 2006.

20 \_\_\_\_\_  
21 HOLLY A. SCHMID  
22 Notary Public -- CSR  
23 084-98-254587  
24